

Mechanochemical preparations of Donor/Acceptor materials based on coronene and silver(I) pyrazolate metallacycles.

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NMR studies

Table S1. A selection of ^1H NMR chemical shifts of adducts **1-3** and of the starting compounds. Δ_{ppm} is the difference of the chemical shift of the hydrogen $\text{C}_4\text{-H}$ of the pyrazole of the CTC upon stacking with the Coronene (third column) and of the terminal hydrogen of coronene upon stacking with the CTC (last column). Data are relative to ^1H NMR spectra acquired in DMSO^{a} or $\text{CD}_2\text{Cl}_2^{\text{b}}$.

Compounds	^1H NMR CTCs $\text{C}_4\text{-H}$ resonance (ppm)	Δ_{ppm} CTCs $\text{C}_4\text{-H}$ resonance (ppm)	^1H NMR Coronene C-H resonance (ppm)	Δ_{ppm} Coronene C-H resonance (ppm)
Coronene	-	-	9.10 ^a ; 8.98 ^b	-
$[\text{pz}^{\text{CF}_3}\text{Ag}]_3$	7.20 ^a	-	-	-
$[\text{pz}^{\text{NO}_2}\text{Ag}]_3$	7.56 ^a	-	-	-
$[\text{pz}^{\text{CF}_3}\text{Cu}]_3$	7.07 ^b	-	-	-
$\{[3,5\text{-(CF}_3)_2\text{pz}]\text{Ag}\}_3@$ Coronene (1)	7.10 ^a	-0.10	9.10 ^a	0
$\{[3,5\text{-(NO}_2)_2\text{pz}]\text{Ag}\}_3@$ Coronene (2)	7.40 ^a	-0.16	9.10 ^a	0
$\{[3,5\text{-(CF}_3)_2\text{pz}]\text{Cu}\}_3@$ Coronene (3)	7.04 ^b	-0.03	8.95 ^b	-0.03

IR spectroscopy studies

Table S2. Main IR absorptions for adducts **1-3** and relative vibrational modes; notes on the redshifts or blueshifts, on comparison with the absorptions of the coronene and the corresponding starting CTC.

$\{[3,5-(CF_3)_2pz]Ag\}_3 @$ Coronene (cm ⁻¹) (1)	Coronene (cm ⁻¹)	$\{[3,5-(CF_3)_2pz]Ag\}_3$ (cm ⁻¹)	Main Vibrational mode	Reference	notes
3149		3153	C ₄ -H stretching	[1]	REDSHIFT
3063	3047		Aromatic C-H stretching	[2]	BLUESHIFT
3028	3018			[2]	BLUESHIFT
1619	1615		Aromatic C-C stretching	[2]	BLUESHIFT
1610	1607		Aromatic C-C stretching	[2]	BLUESHIFT
1501		1505	Aromatic C=N stretching	[1]	REDSHIFT
1315	1312		Aromatic C-C stretching	[2]	BLUESHIFT
1126		1143	C-F stretching	[3]	REDSHIFT
861	847		In-plane C-H bending	[2]	BLUESHIFT

$\{[3,5-(CF_3)_2pZ]Cu\}_3$ @ Coronene (cm ⁻¹) (3)	Coronene (cm ⁻¹)	$\{[3,5-(CF_3)_2pZ]Ag\}_3$ (cm ⁻¹)	Main Vibrational mode	Reference	notes
3149		3163	C ₄ -H stretching	[1]	REDSHIFT
3061	3048		Aromatic C-H stretching	[2]	BLUESHIFT
3026	3018			[2]	BLUESHIFT
1618	1614		Aromatic C-C stretching	[2]	BLUESHIFT
1610	1607			[2]	BLUESHIFT
1505		1508	Aromatic C=N stretching	[1]	REDSHIFT
1315	1312		Aromatic C-C stretching	[2]	BLUESHIFT
1123		1126	C-F stretching	[3]	REDSHIFT
968	957		C-H bending	[2]	BLUESHIFT
861	847		In-planes C-H bending	[2]	BLUESHIFT

Table S3 EDX analysis of the surface of the samples treated with solventless mechanochemical grinding top: Spot 1; bottom: Spot 2.

Element	At. No.	Net	Mass [%]	Norm. Mass [%]	Atom [%]	abs. error [%] (1 sigma)	abs. error [%] (2 sigma)	abs. error [%] (3 sigma)
Carbon	6	133944	88.33	88.33	91.57	9.34	18.68	28.02
Oxygen	8	6523	9.62	9.62	7.49	1.23	2.45	3.68
Fluorine	9	1132	1.16	1.16	0.76	0.21	0.42	0.63
Silicon	14	253	0.24	0.24	0.11	0.03	0.07	0.10
Silver	47	269	0.65	0.65	0.07	0.05	0.10	0.15
			Sum 100.00	Sum 100.00	Sum 100.00			
Element	At. No.	Net	Mass [%]	Norm. Mass [%]	Atom [%]	abs. error [%] (1 sigma)	abs. error [%] (2 sigma)	abs. error [%] (3 sigma)
Carbon	6	46475	26.86	37.51	59.43	2.96	5.92	8.88
Nitrogen	7	2383	3.86	5.39	7.33	0.57	1.14	1.71
Oxygen	8	3847	4.05	5.66	6.73	0.56	1.12	1.68
Fluorine	9	18682	14.79	20.65	20.69	1.72	3.44	5.16
Silicon	14	601	0.57	0.80	0.54	0.05	0.10	0.15
Silver	47	10382	21.48	29.99	5.29	0.81	1.62	2.42
			Sum 71.61	Sum 100.00	Sum 100.00			

Table S4 EDX analysis of the surface of the samples treated with LAG (Liquid Assisted Grinding) top: Spot 3; bottom: Spot 4.

Element	At. No.	Net	Mass [%]	Norm. Mass [%]	Atom [%]	abs. error [%] (1 sigma)	rel. error [%] (1 sigma)
Carbon	6	39769	26.21	32.65	51.65	2.91	11.10
Nitrogen	7	2463	3.85	4.80	6.51	0.57	14.68
Oxygen	8	8320	8.45	10.53	12.50	1.06	12.49
Fluorine	9	19762	15.98	19.91	19.91	1.85	11.58
Aluminium	13	1256	0.64	0.80	0.56	0.05	8.33
Silicon	14	6421	5.37	6.69	4.53	0.24	4.56
Silver	47	10430	19.77	24.63	4.34	0.75	3.77
			Sum 80.26	Sum 100.00	Sum 100.00		
Element	At. No.	Net	Mass [%]	Norm. Mass [%]	Atom [%]	abs. error [%] (1 sigma)	rel. error [%] (1 sigma)
Carbon	6	36029	25.08	31.94	50.49	2.80	11.15
Nitrogen	7	2205	3.56	4.54	6.15	0.53	14.96
Oxygen	8	8918	9.29	11.83	14.04	1.15	12.39
Fluorine	9	17903	15.04	19.15	19.14	1.75	11.66
Aluminium	13	1450	0.79	1.01	0.71	0.06	7.60
Silicon	14	7190	6.15	7.83	5.30	0.28	4.50
Silver	47	9706	18.61	23.69	4.17	0.70	3.78
			Sum 78.53	Sum 100.00	Sum 100.00		

Figure S1. Overlapped ^1H NMR spectra of $\{[3,5-(\text{CF}_3)_2\text{pz}]\text{Ag}\}_3@$ Coronene (**1**) (blue line), $\{[3,5-(\text{CF}_3)_2\text{pz}]\text{Ag}\}_3$ (red line), and coronene (green line).

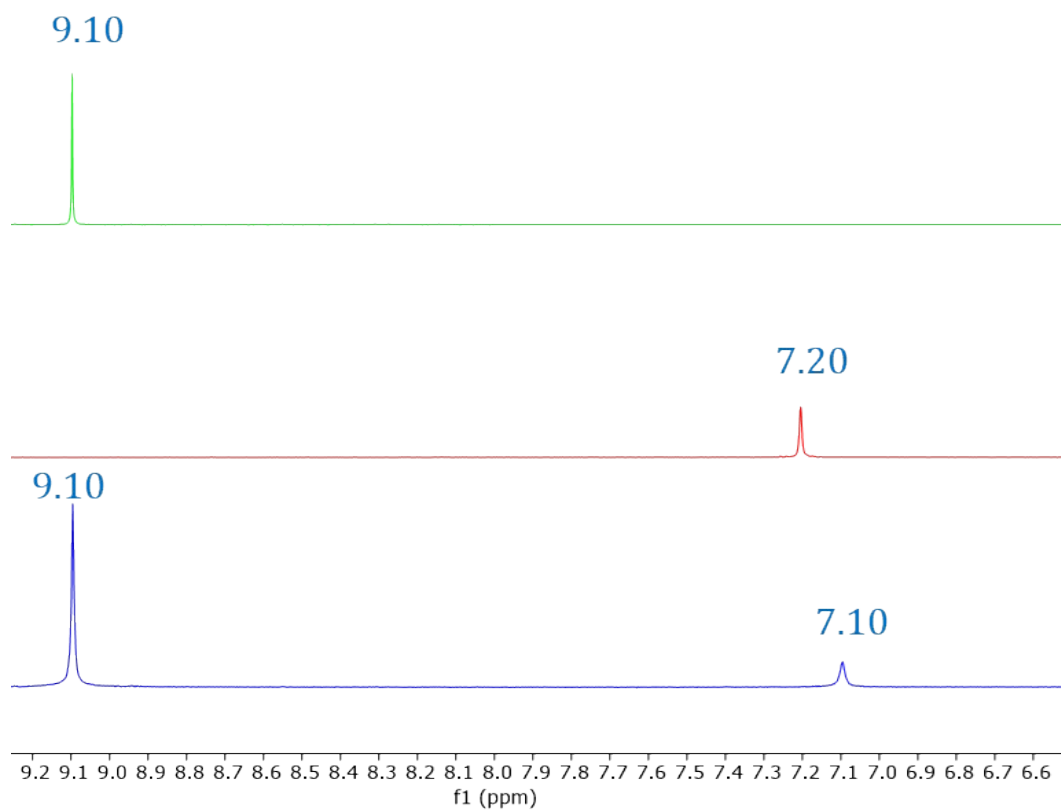


Figure S2. Overlapped ^1H NMR spectra of $\{[3,5-(\text{NO}_2)_2\text{pz}]\text{Ag}\}_3@$ Coronene (**2**) (orange line), $\{[3,5-(\text{NO}_2)_2\text{pz}]\text{Ag}\}_3$ (blue line), and coronene (red line).

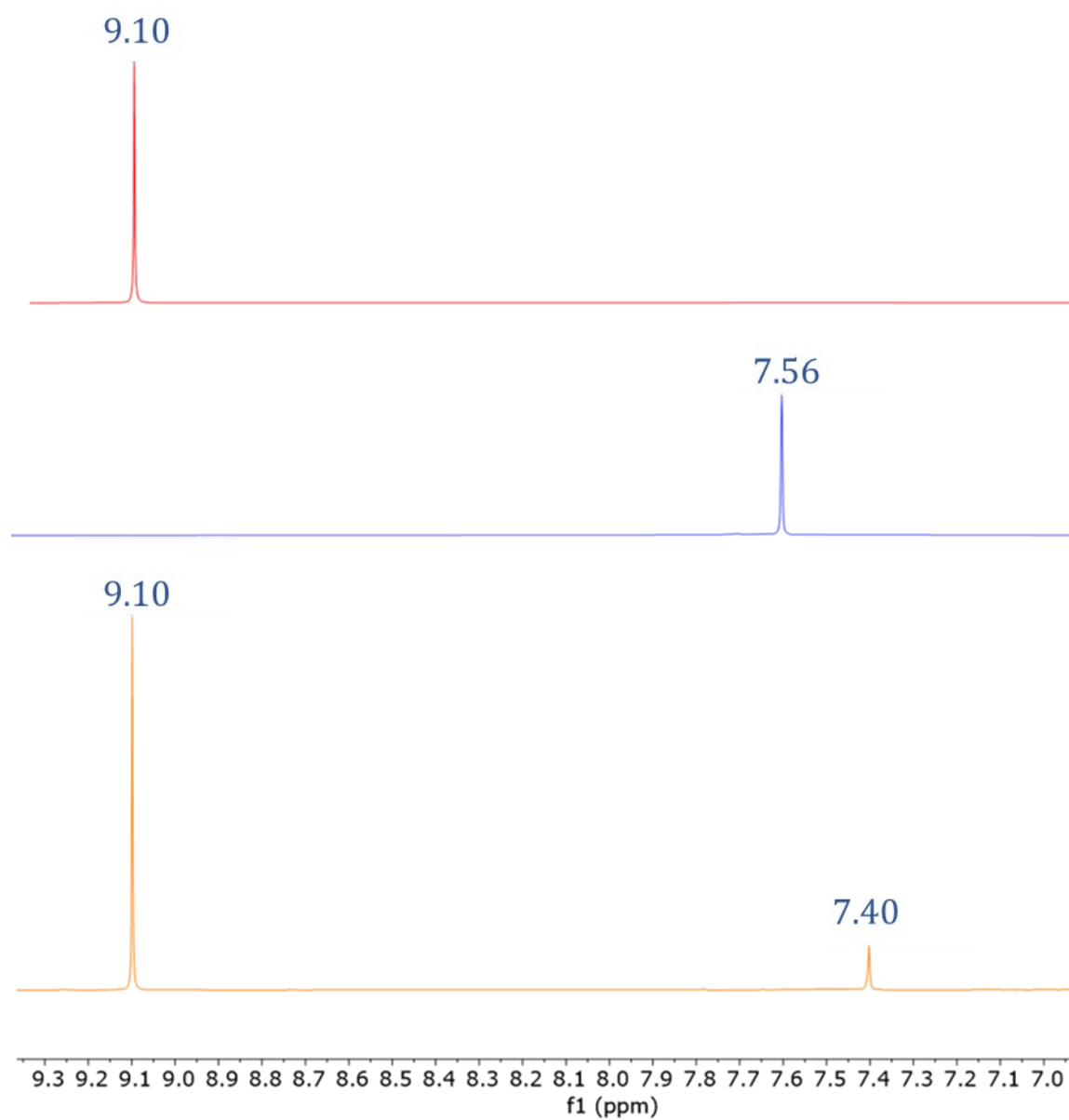


Figure S3. Overlapped ^1H NMR spectra of $\{[3,5-(\text{CF}_3)_2\text{pz}]\text{Cu}\}_3@$ Coronene (**3**) (orange line), $\{[3,5-(\text{CF}_3)_2\text{pz}]\text{Cu}\}_3$ (blue line), and coronene (red line).

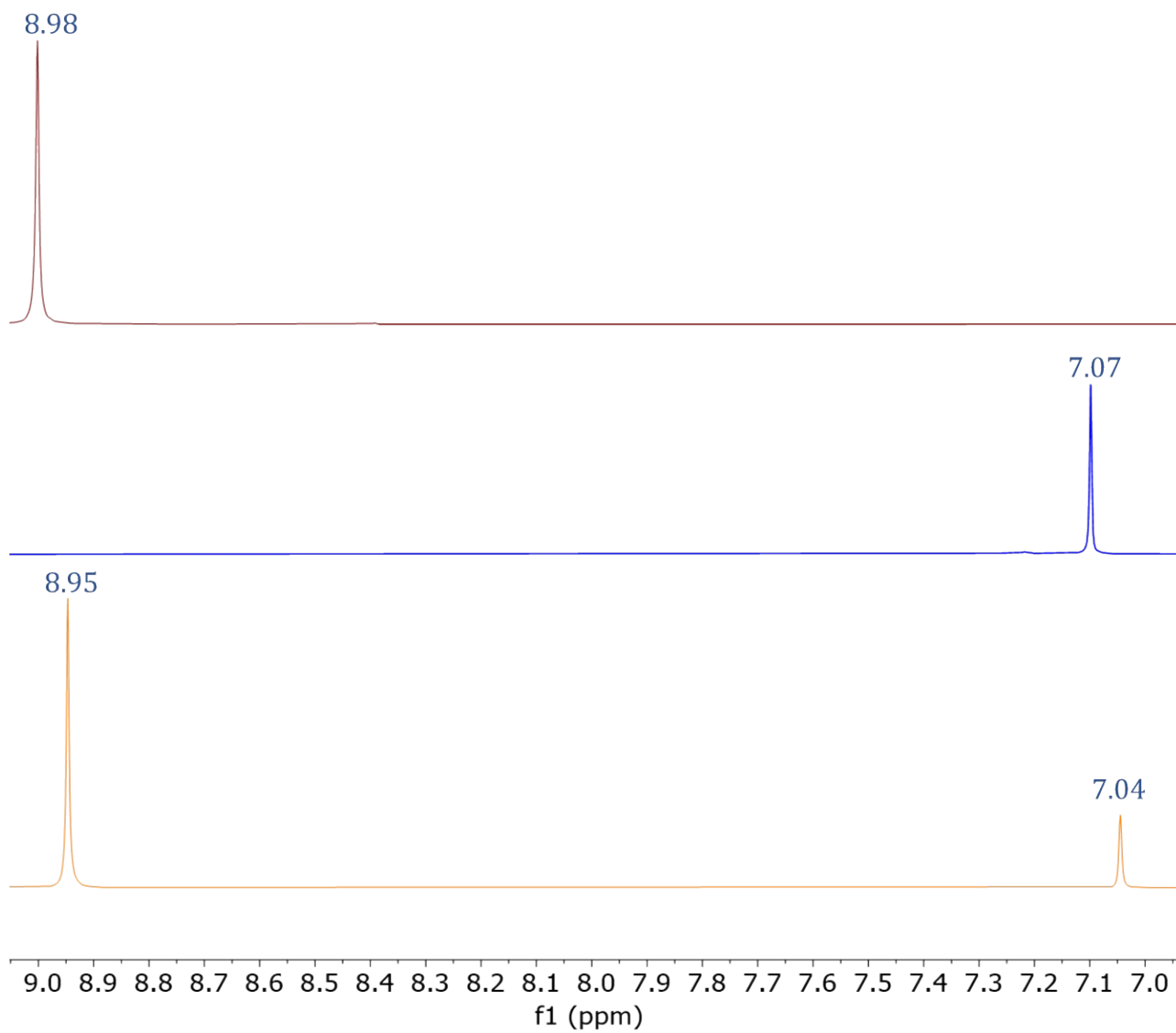
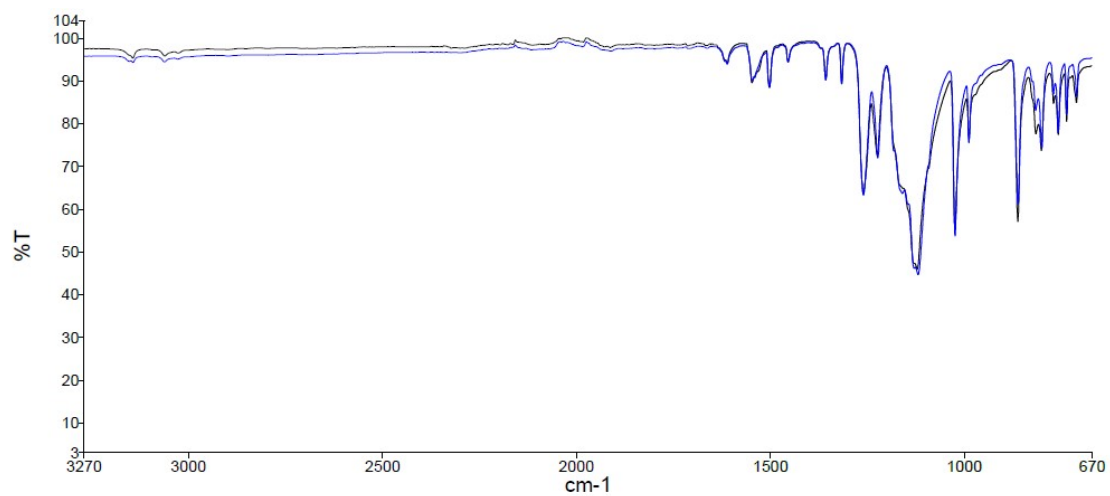
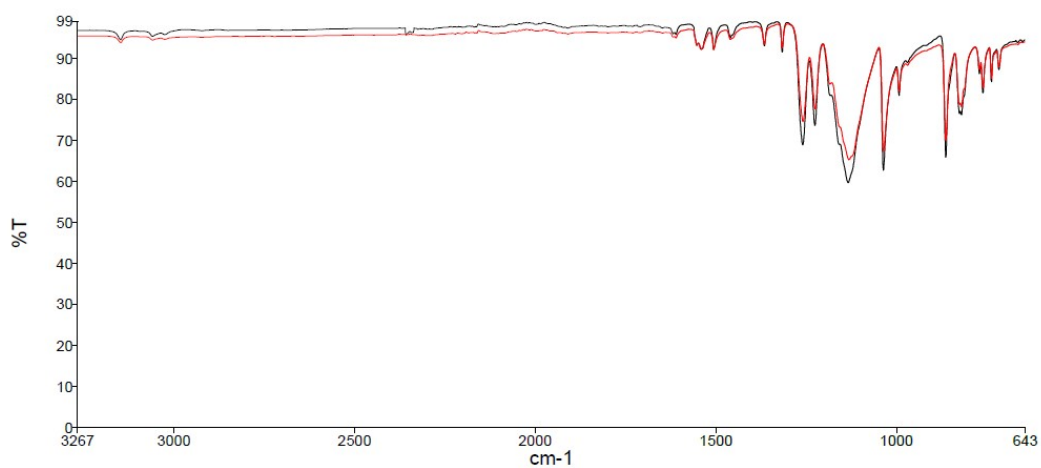


Figure S4. Superimposition of IR spectra recorded for the microcrystalline products **1** (a), **2** (b) and **3** (c) obtained by solvent-mediated syntheses and 10 minutes LAG (liquid-assisted grinding), recorded in the range 3200-700 cm^{-1} .

a)



b)



c)

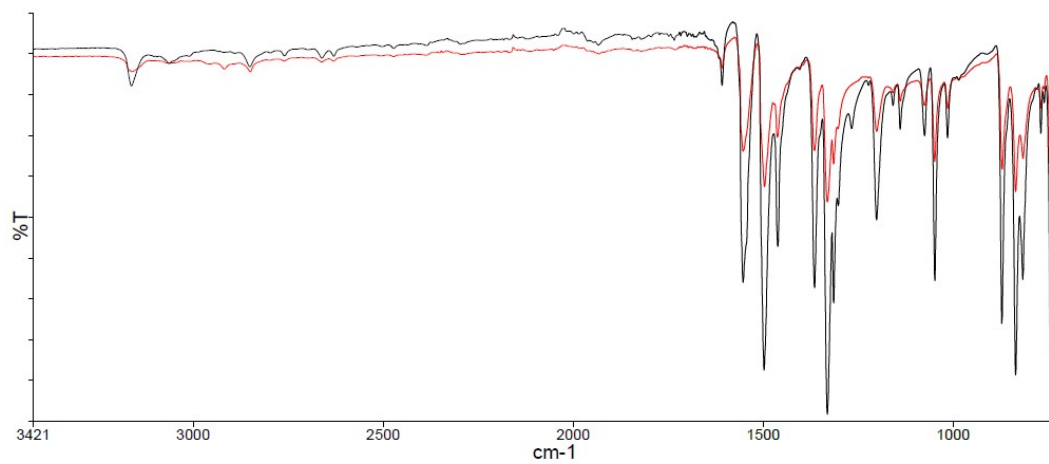
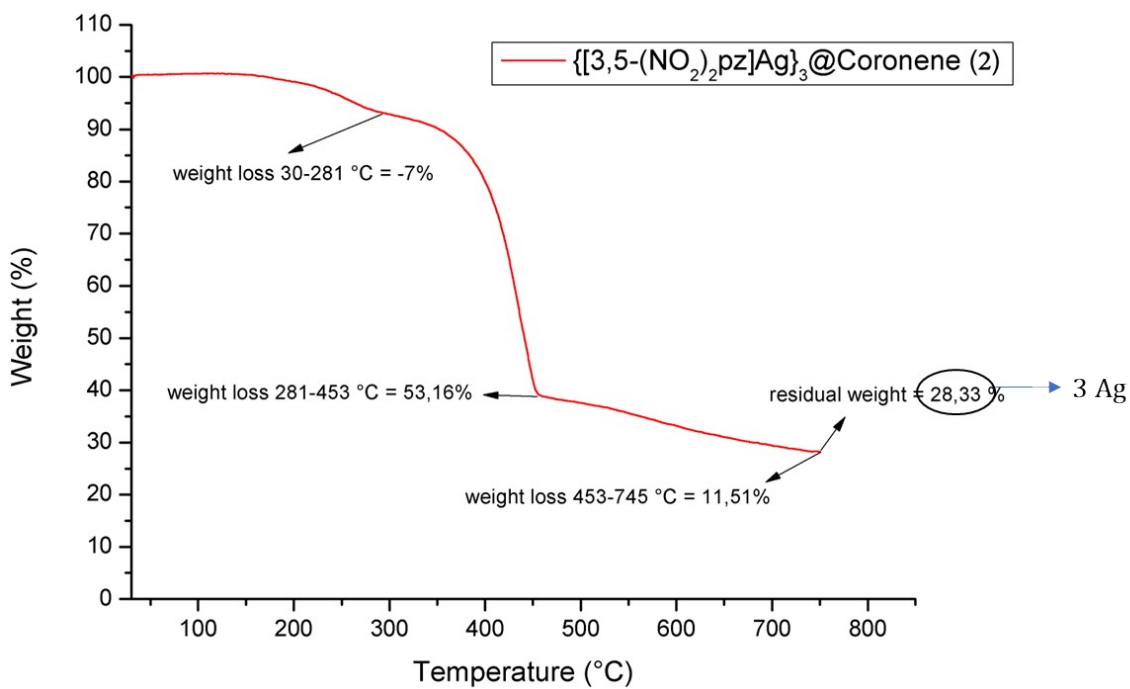
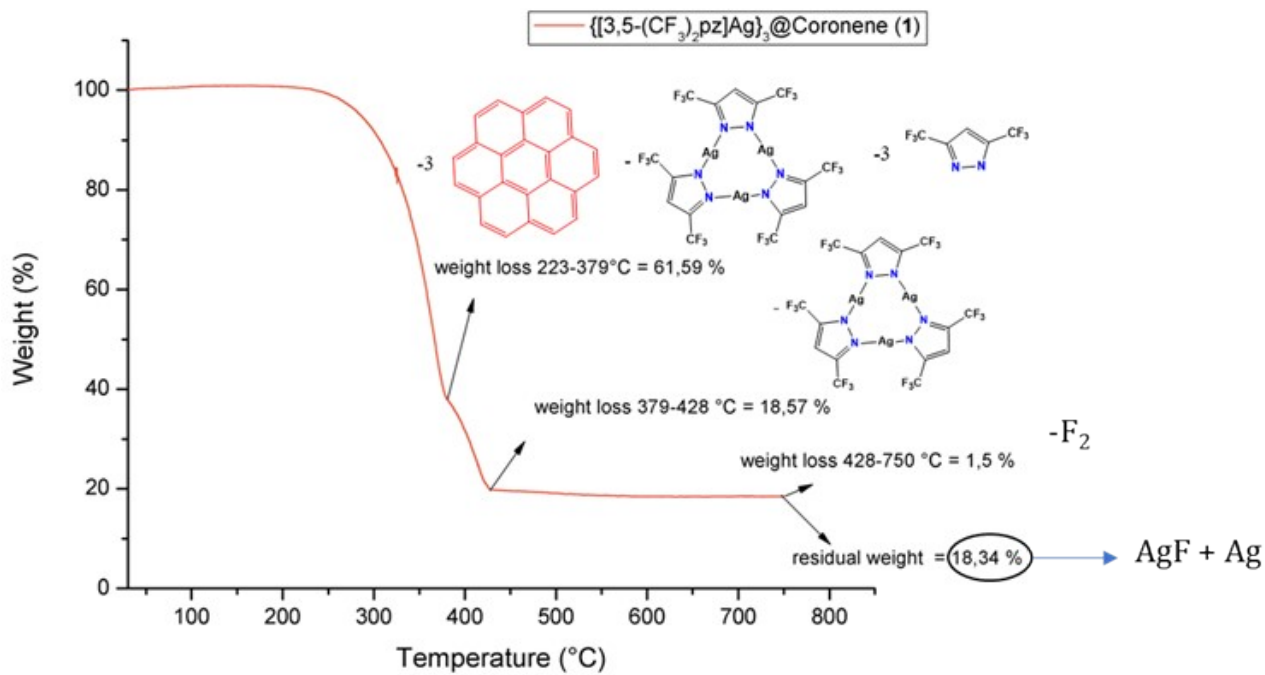


Figure S5. TGA plot with decomposition patterns of 1,2, and 3 (from top to bottom).



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Figure 5S. Continue.

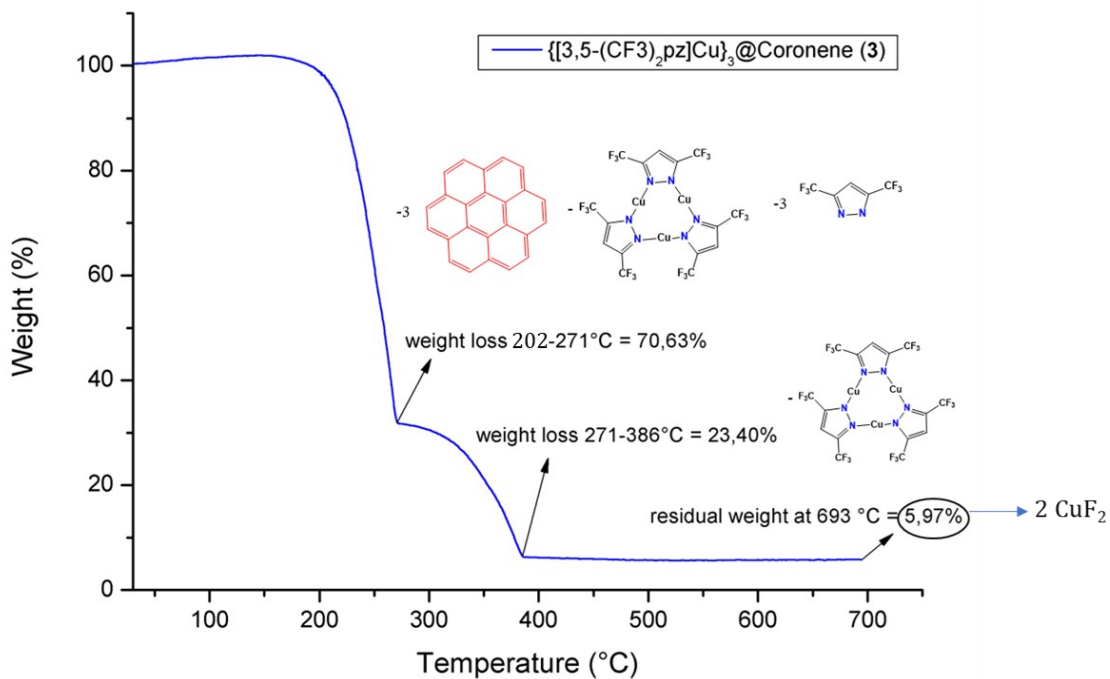
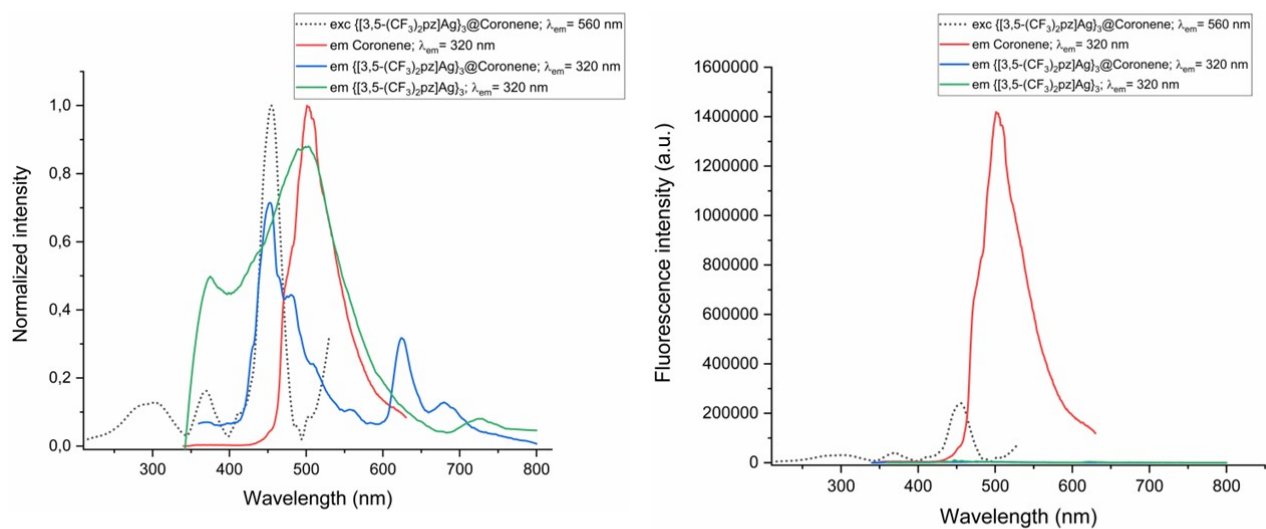


Figure S6. Normalized (left) and absolute (right) solid-state excitation/emission spectra of the adduct $\{[3,5-(\text{CF}_3)_2\text{pz}]\text{Ag}\}_3@ \text{Coronene}$ **1**, a), and $\{[3,5-(\text{NO}_2)_2\text{pz}]\text{Ag}\}_3@ \text{Coronene}$ **2**, b) in comparison with the emission spectra of the starting compounds in the range 300-630 nm.

a)



b)

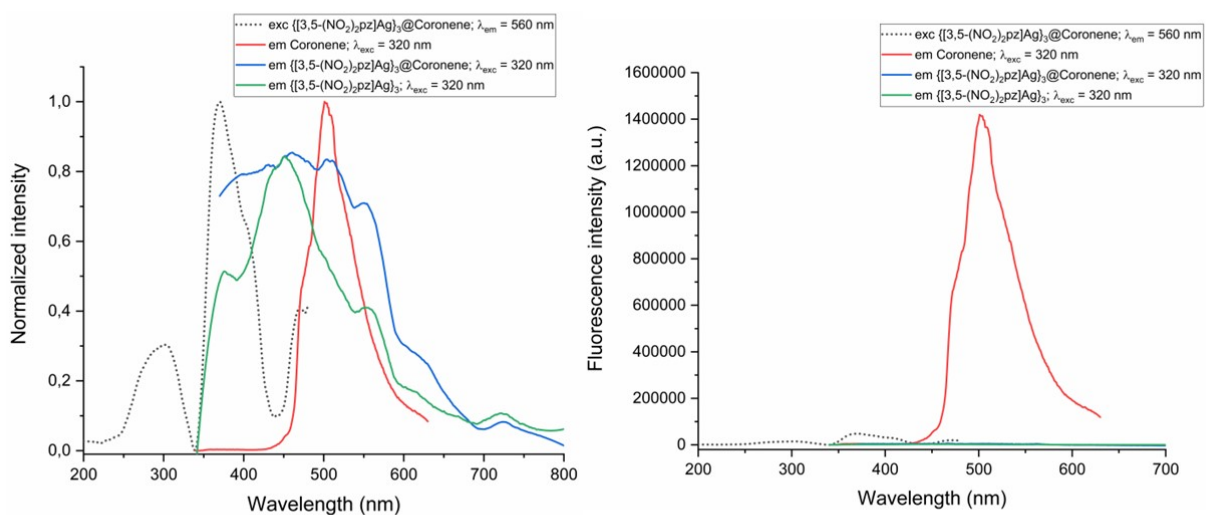


Figure S7. Solid state emission spectra of the coronene, of the $[(CF_3)_2pzCu]_3$ and of the adduct **3** upon excitation at 320 nm at room temperature.

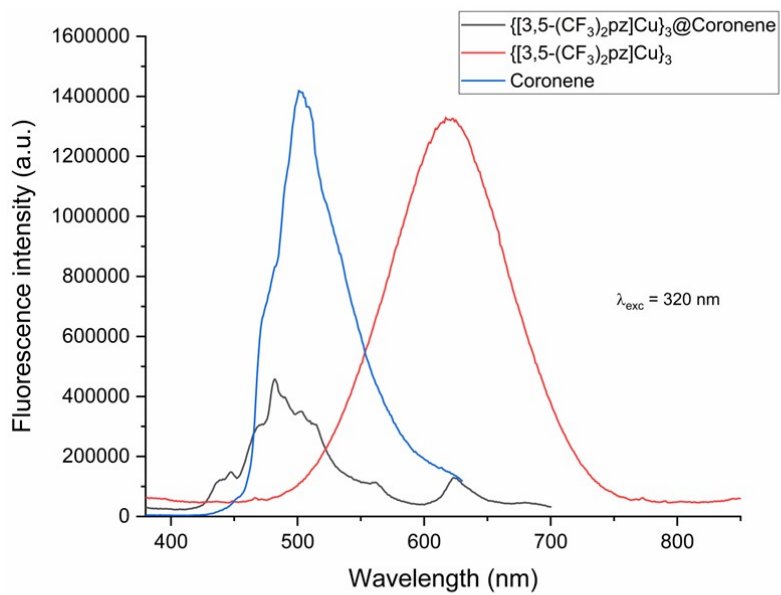


Figure S8. Normalized solid state excitation/emission spectra of the coronene/ $[(CF_3)_2pzCu]_3$, adduct **3**, upon excitation at 320 nm, 360 nm and 400 nm. Upon excitation at 320 nm a weak emission at 560 nm is observed.

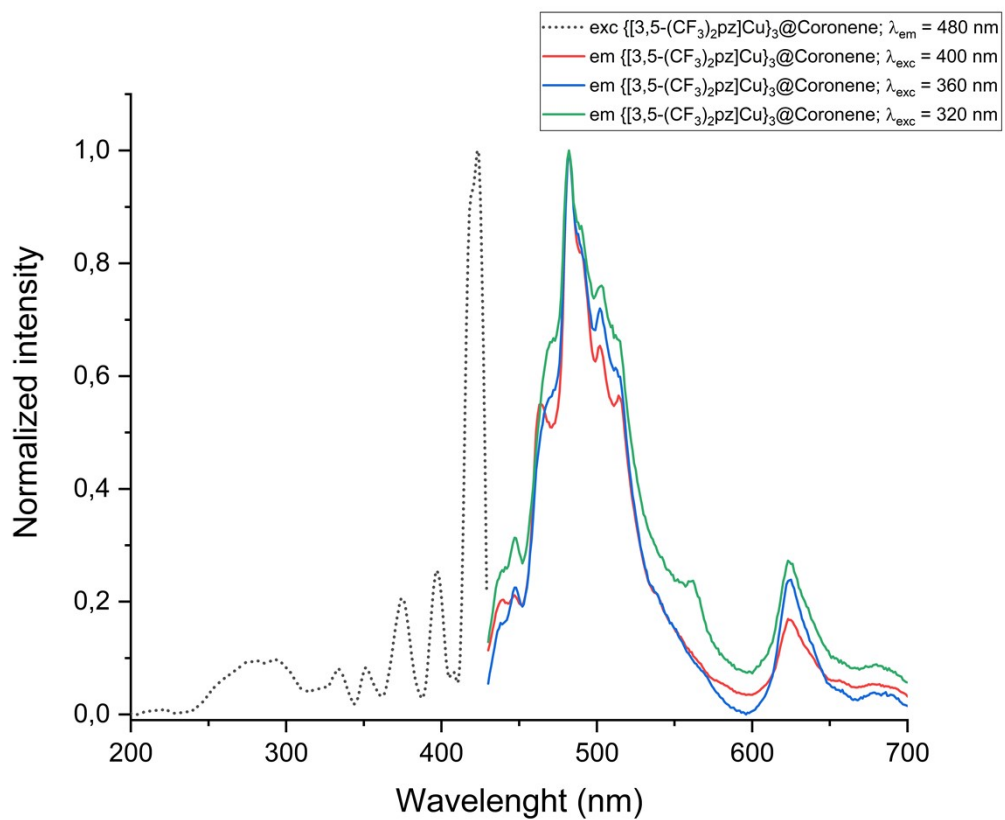


Figure S9. Excitation and emission spectra for $[(CF_3)_2pzCu]_3$, top left, for coronene, top right, for $[(NO_2)_2pzAg]_3$, bottom left, and for $[(CF_3)_2pzAg]_3$, bottom right.

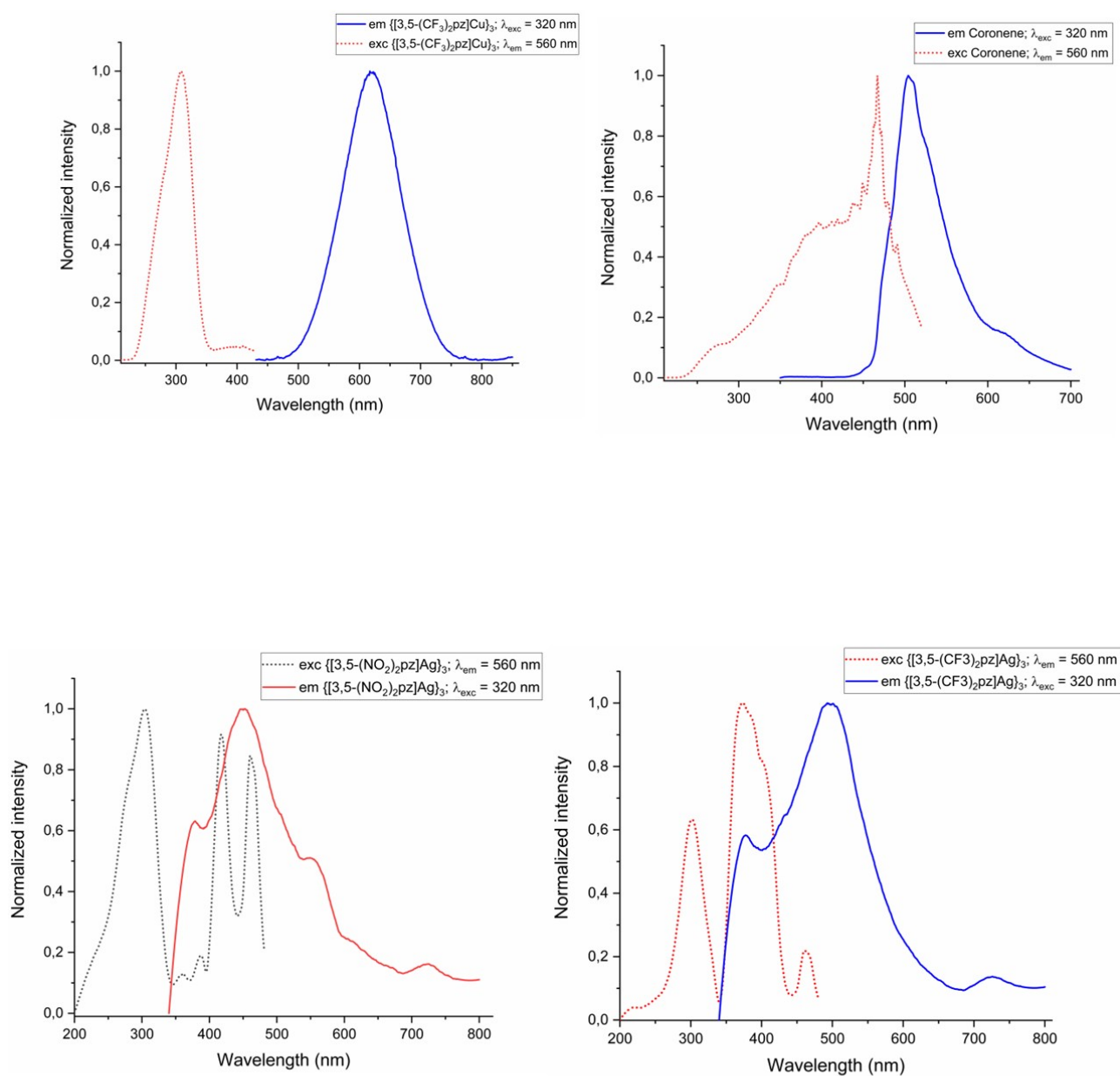


Figure S10. Emission bands obtained by exciting at 350 nm a 24.26 μM solution of coronene in DCM and upon titrating with a solution of $\{[(\text{CF}_3)_2\text{pzCu}]_3\}$ in DCM. The addition of the CTC was performed till a ratio of coronene: $\{[(\text{CF}_3)_2\text{pzCu}]_3\}$ almost 1:2 (first addition mole ratio $\{[(\text{CF}_3)_2\text{pzCu}]_3\}$ /coronene = 0.269). The adding of $\{[(\text{CF}_3)_2\text{pzCu}]_3\}$ did not affected the intensity and emission maxima of coronene.

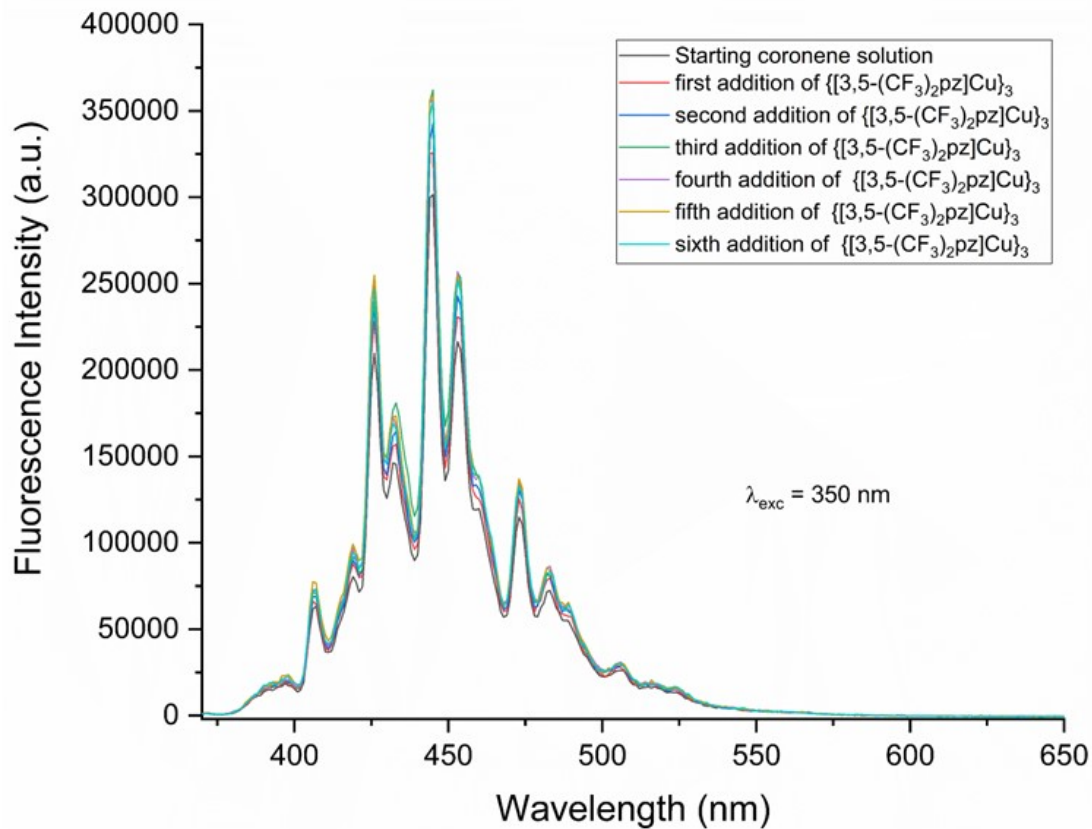


Figure S11. Tauc plots of $\{[3,5-(CF_3)_2pz]Ag\}_3$ with direct (left) and indirect (right) band gap calculations.

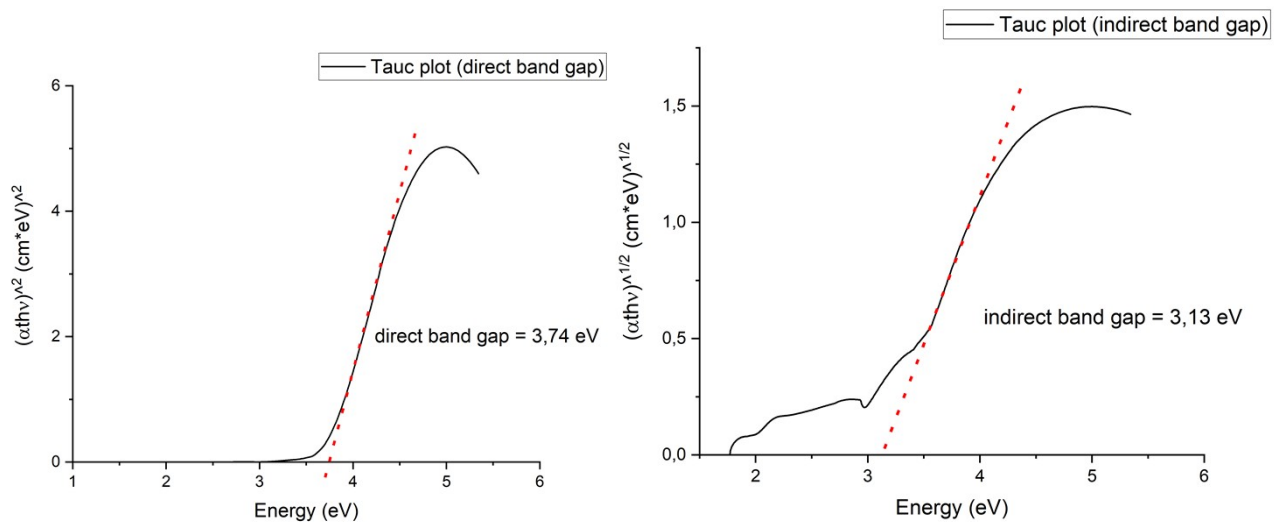
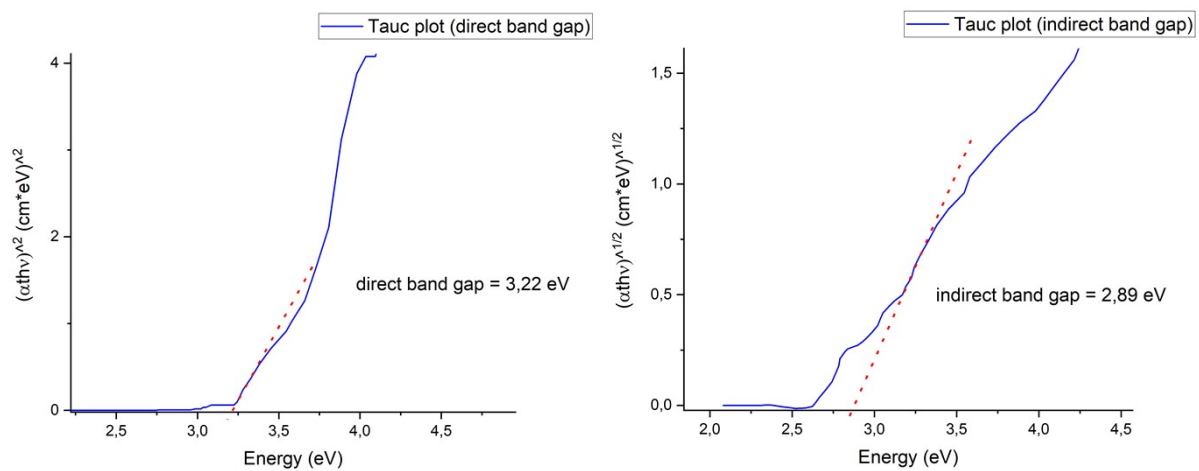


Figure S12. Tauc plot of coronene with direct (left) and indirect (right) band gap calculations.



References:

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