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Coffee silverskin: characterization of B-vitamins, macronutrients, minerals and phytosterols

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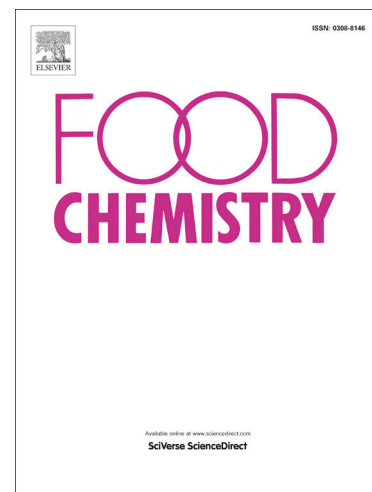
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1 **Coffee silverskin: characterization of B-vitamins, macronutrients, minerals and**
2 **phytosterols**

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Abstract

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27 The present study assessed the nutritional composition of coffee silverskin (CSS) obtained from
28 arabica roasted coffee. Following validated analytical methods, CSS resulted to be a high source of
29 proteins (14.2 g/100 g) and dietary fibers (51.5 g/100 g). Moreover, the mineral analysis revealed
30 high contents of calcium (1.1 g/100 g) and potassium (1.0 g/100 g). To date, this study provided the
31 widest mineral profile of CSS with 30 minerals targeted including 23 microminerals with high
32 levels of iron (238.0 mg/kg), manganese (46.7 mg/kg), copper (37.9 mg/kg), and zinc (31.9 mg/kg).
33 Moreover, vitamins B₂ (0.18–0.2 mg/kg) and B₃ (2.5–3.1 mg/kg) were studied and reported for the
34 first time in CSS. β -sitosterol (77.1 mg/kg), campesterol, stigmasterol, and Δ 5-avenasterol, were
35 also observed from the phytosterol analysis of CSS with a total level of 98.4 mg/kg. This rich
36 nutritional profile highlights the potential values of CSS for innovative reuses in bioactive
37 ingredients development.

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49 **Keywords:** Coffee silverskin, coffee by-product, nutrients, B-vitamins, phytosterols, resilience

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52 1. Introduction

53 Coffee silverskin (CSS) is the integument of coffee beans, which is released as the main by-product
54 of the coffee roasting process. Considering the high production and consumption of coffee,
55 thousands of tonnes of CSS are produced each year in coffee-roasting industries worldwide (Juan-
56 García et al., 2021). Indeed, according to the International Coffee Organization (ICO) statistics,
57 around 9.9 billion kg of coffee are yearly consumed (ICO, 2021), corresponding to 200–400 million
58 kg of CSS generated (Barbero-López et al., 2020; del Pozo et al., 2020). Moreover, according to
59 ~~FAO, growth projections on coffee production and consumption are expected to increase over the~~
60 ~~next years. In 2019 the world production accounted for 10,035,576 tonnes with a harvested area of~~
61 ~~11,120,498 hectares, and data compared to 2009 respectively account for + 1.5% in production and~~
62 ~~+ 6.31% in the harvesting surface (FAO, 2019).~~

63 At the global level, the significant amount of discards in the coffee industry (23 million tons of
64 waste per year) is leading to a greater and greater valorization of such food wastes, becoming a
65 priority research line to achieve a more sustainable food value-chain, in line with the growing
66 attention of consumers and the international community to sustainability and resilience in the agri-
67 food sector (Conrad, Tichenor, and Blackstone, 2021).

68 ~~Wastes generated from the coffee industries have become a resource, especially in agricultural~~
69 ~~contexts. For example, spent coffee grounds are used with a «Zero Wastes» approach with different~~
70 ~~applications in agriculture, such as fertilizer for growing mushrooms, as they contain minerals and~~
71 ~~nutrients that are useful for their growth. Other applications contemplate the use of coffee wastes to~~
72 ~~create pellet or to produce biobased materials such as pens and cups.~~

73 ~~In addition, therefore~~ various studies have been carried out to develop innovative solutions and
74 propose value-added applications for the valorization of CSS according to principles of the green
75 and circular economy. The most recent studies on CSS valorization described the conversion of
76 CSS into different value-added products including biopolymers composite for packaging
77 development (Garcia & Kim, 2021), feedstock for wood preservative formulation (Barbero-López

78 et al., 2020), ingredients for functional foods formulation (Gemechu, 2020), and functional
79 ingredients in cosmetic products (dos Santos et al., 2021).

80 These applications can be performed thanks to the chemical characteristics and composition of CSS.
81 Indeed, CSS is reported to be a high source of dietary fibers and phytochemicals including
82 chlorogenic acids, melanoidins, and caffeine (Castaldo et al., 2020). Various studies assessed the
83 bioactive compounds and the nutritional composition of CSS giving more attention to the fibers
84 composition, fat profile, proteins, and polyphenols content (del Pozo et al., 2020). Other classes of
85 bioactive compounds in CSS have not been comprehensively assessed including minerals, vitamins,
86 and phytosterols.

87 The mineral profile of CSS has been discussed in few studies focusing principally on the most
88 important macrominerals (Wen et al., 2020). Nevertheless, microminerals and toxic heavy metals,
89 which are health-impacting elements have not been considered. Similarly, phytosterols, which are
90 cholesterol-lowering bioactive compounds have been poorly assessed in CSS. Indeed, to the
91 author's knowledge, the characterization and the content of phytosterols in CSS have been
92 performed in just one study (Toschi et al., 2014). However, the sample studied was obtained from a
93 mixture of arabica and robusta coffee. Considering vitamins, the related literature is scarce. The
94 vitamin E profile of CSS has been investigated by Costa et al., 2018 (4.3 mg/100 g) and Bessada et
95 al., 2018 (5.3–16.8 mg/100 g). Besides, B-vitamins, which some of them derive from the thermal
96 degradation of trigonelline could be abundant micronutrients in CSS. Indeed, trigonelline is an
97 important alkaloid in coffee (averagely 1 g/100 g DW) that can be highly degraded during roasting
98 (50–80% of decomposition) (Gemechu, 2020). However, to date, no study has assessed the profile
99 of B-vitamins in CSS.

100 Therefore, the present study aims to perform a comprehensive analysis of nutrients, microelements,
101 and bioactive compounds in CSS. The analyses will be focused on non-assessed and poorly studied
102 compounds (minerals, B-vitamins i.e., riboflavin and niacin, and phytosterols) in order to propose
103 novel value-added applications to CSS.

104 **2. Materials and methods**

105 **2.1 Chemicals and reagents**

106 Riboflavin (purity > 98%, C₁₇H₂₀N₄O₆, molecular weight 376.4 g/mol, CAS Number 50-81-7),
107 nicotinamide (purity > 99.5%, C₆H₆N₂O, molecular weight 122.1 g/mol, CAS Number 98-92-0),
108 nicotinic acid (purity > 99.5%, C₆H₅NO₂, molecular weight 123.1 g/mol, CAS Number 59-67-6),
109 and β-sitosterol (C₂₉H₅₀O, molecular weight 414.7 g/mol, CAS Number 83-46-5) were purchased
110 from Sigma-Aldrich (Milan, Italy). Campesterol (C₂₈H₄₈O, molecular weight 400.7 g/mol, CAS
111 Number 474-62-4), Δ⁵-avenasterol (C₂₉H₄₈O, molecular weight 412.7 g/mol, CAS Number 17605-
112 67-3), and stigmasterol (C₂₉H₄₈O, molecular weight 412.7 g/mol, CAS Number 83-48-7) were
113 provided by Phytolab (Vestenbergsgreuth, Germany). Individual stock solutions of nicotinamide
114 and nicotinic acid were prepared by dissolving 10 mg of pure powder into 10 mL of acetic acid 2%
115 (v/v) solution, while stock solutions of phytosterol standards were prepared in chloroform (10
116 mg/ml). On the other hand, the stock solution of riboflavin was obtained by dissolving 4 mg of the
117 pure standard into 100 mL of acetic acid solution. All stock solutions were stored in a glass-
118 stoppered bottle at 4 °C in the dark and different solution concentrations of the three analytes
119 (standard working solutions) were daily prepared by diluting appropriately the stock solutions in the
120 acetic acid solution for B-vitamins, and methanol for phytosterols. Ultra-pure water was obtained
121 using a Milli-Q SP Reagent Water System (Millipore, Bedford, MA, USA) while HPLC-grade
122 acetonitrile was purchased from Sigma-Aldrich (Milan, Italy). HPLC-grade formic acid 99–100%
123 was bought from J.T. Baker B.V. (Deventer, Holland). All other chemicals were analytical grade.
124 Before liquid chromatography-mass spectrometry (LC-MS) analysis all samples were filtered with
125 Phenex™ RC 4 mm 0.2 μm syringeless filter, Phenomenex (Castel Maggiore, Italy).

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128 **2.2 Coffee silverskin collection and preparation**

129 Coffee silverskin (CSS), which was collected after roasting green beans (*Coffea arabica* L.) at
130 medium/dark level, was kindly provided by Illycaffè S.p.A. (Trieste, Italy). Before extraction and
131 analysis, CSS was frozen with liquid nitrogen and then milled with an Ariete Blendy 570 grinder
132 (Florence, Italy) to obtain a homogeneous powder that was stored at 4 °C, at dark, until use.

133 **2.3 Macronutrient and fiber analysis** ~~Nutritional analysis and mineral content~~

134 CSS samples were analyzed to assess their chemical composition through the AOAC procedures.
135 The moisture content was determined by drying the sample in the oven (24 h, 133 °C) until steady
136 weight. The crude fat level was obtained by the Soxhlet extraction of a fixed quantity of CSS
137 sample with petroleum ether; the protein content was estimated through the Kjeldahl method; the
138 ash content was determined by incineration at 600 ± 15 °C; dietary fiber levels were evaluated
139 through a gravimetric method after acidic hydrolysis, while total carbohydrates were calculated by
140 the difference.

141 **2.4 Mineral analysis**

142 The mineral composition was performed by the digestion of dried samples with nitric acid
143 (Suprapur, Merck) followed by analysis with inductively coupled plasma mass spectrometry (ICP-
144 MS, Agilent Technologies, Santa Clara, CA) (Nkuimi et al. 2019).

145 **2.5 Vitamin B₂ and B₃ analysis** ~~2.4 Vitamin B₂ and B₃ extraction and LC-MS quantification~~

146 The extraction of vitamin B₂ (riboflavin) and B₃ (niacin, i.e., nicotinic acid and nicotinamide) was
147 performed taking into consideration a paper by Caprioli, Sagratini, Vittori, and Torregiani (2018),
148 who evaluated the efficiency of different extraction procedures for these vitamins. For the current
149 research was selected only the most efficient procedure obtained in that research, in terms of
150 recoveries, quantitative results obtained, and rapidity but varying the solvent-to-sample ratio (SSR)
151 such as 20:1, 30:1, and 40:1 (v/w, mL/g). The extraction of nicotinamide, nicotinic acid, and
152 riboflavin was carried out by adding 1 g of CSS powder together with 0.1 M hydrochloric acid
153 (HCl) into a 100 mL flask and the hydrolysis was performed at 100 °C for 30 min. Samples were
154 then cooled down at room temperature and were filtered with Whatman filter paper; the solution pH

155 was later adjusted to 4.0–4.5 by adding an appropriate amount of 2 M sodium acetate. Finally, it
156 was diluted to 50 mL with ultra-pure water, and before high-performance liquid chromatography-
157 tandem mass spectrometry (HPLC-MS/MS) analysis, the sample was centrifuged at 13,000 rpm for
158 10 min and filtered with a 0.2 µm membrane filter.

159 The quantification of niacin and riboflavin was performed following a previous procedure (Caprioli
160 et al., 2018). Briefly, an Agilent 1290 Infinity series and a Triple Quadrupole 6420 from Agilent
161 Technology (Santa Clara, CA) were employed for the current research. The MS system was
162 composed of an electrospray ionization (ESI) source which operated in positive ionization mode.
163 The separation of the target compounds was achieved using a Kinetex HILIC analytical column
164 (100 mm× 4.6 mm i.d., particle size 2.6 µm) from Phenomenex (Torrance, CA, USA). The mobile
165 phase was composed of water (A) and acetonitrile (B) both with 0.1% of formic acid and the flow
166 rate was 0.8 mL/min. The elution was achieved in isocratic condition and the injection volume was
167 2 µL. The source parameters were as following: temperature of the drying gas, 300 °C, gas flow, 12
168 L/min, nebulizer pressure, 50 psi, and capillary voltage, 4,000 V. The acquisition was performed in
169 “Multiple Reaction Monitoring” (MRM) mode and the most abundant transitions were used for
170 quantitation while the other for confirming the analyte presence. The HPLC-MS/MS acquisition
171 parameters including the selected mass transitions and retention time (Rt) are shown in **Table 1**.

172 **2.5.1 Method validation**

173 The analysis of vitamin B₂ (riboflavin) and B₃ (nicotinic acid and nicotinamide) has been carried
174 out following a developed method (Caprioli et al., 2018) which used a hydrophilic interaction
175 chromatography (HILIC) instead of classic reversed-phase (RP) since HILIC is more suitable for
176 polar compounds as hydrosoluble vitamins are, and it enhances compatible with MS system (Porter,
177 & Lodge, 2021). Before the analysis of CSS, the analytical procedure has been validated by
178 assessing the linearity, sensitivity, repeatability, and recovery (**Table 2**). The linearity was
179 evaluated by injecting eight different concentrations of standard mixtures and preparing calibration

180 curves. These were constructed by plotting the standard peak areas by concentrations and for each
181 analyte, a determination coefficient (R^2) was obtained. All analytes showed good linearity since the
182 R^2 was ≥ 0.9945 . The sensitivity has been studied by measuring the limit of detection (LOD) and
183 limit of quantification (LOQ). LOD and LOQ were estimated by injecting low concentrations of
184 studied compounds and the concentration that generated a signal-to-noise ratio (SNR) of 3 was
185 assigned to LOD while that with SNR of 10 was designated as LOQ. The analytes showed
186 satisfactory sensitivity since LOQ was 3 ng/mL for all vitamins and these values were similar to
187 those reported in the literature (Caprioli et al., 2018). The method repeatability was evaluated by
188 injecting five times three analyte concentrations in the same day and during three consecutive days
189 and the intraday and interday repeatability was calculated and expressed as relative standard
190 deviation (%RSD). The intraday and interday repeatability were 2.5–4.6% and 3.9–6.2%,
191 respectively. Recovery studies were performed at two different spiking levels, 1 and 0.1 $\mu\text{g/g}$ to
192 study the accuracy of the current method. These studies have been evaluated for each SSR level but
193 similar findings have been obtained. **Table 2** shows the recoveries of the procedure which
194 employed SSR of 30:1 and satisfactory recovery levels were found with both spiking levels for all
195 analytes such as 78–86% with 0.1 $\mu\text{g/g}$ and 82–93% with 1 $\mu\text{g/g}$.

196 **2.65 Phytosterols analysis**

197 The analysis of phytosterols (PS) required a prior extraction step, which was performed following
198 the method of Nzekoue et al. (2020). Briefly, 1 g of CSS was submitted to acidic hydrolysis with 4
199 mL of HCl solution (0.25 N) through 10 min of ultrasonication. The extraction was followed by a
200 saponification process at 80 °C for 40 min adding 5 mL of KOH (50 g/100 mL) and 20 mL of
201 ethanol. Then, PS were extracted with 10 mL of hexane three times (10 mL x 3). The collected
202 extract was dried and then derivatized (40 °C, 30 min) with 1 mL of dichloromethane containing 4-
203 dimethylaminopyridine (DMAP) and dansyl chloride (8 mg/mL). Derivatized samples were dried,
204 reconstituted with 1 mL acetonitrile, and filtrated with a 0.45 μm membrane filter before analysis.

205 The identification and quantification of PS were performed through HPLC analyses, using a 1260
206 Infinity LC-DAD system (Agilent Technologies, Santa Clara, CA, USA). Analyses were carried out
207 in isocratic mode with methanol (100%) as mobile phase at a flow rate of 0.5 mL/min. Analytes
208 separation was performed through a Gemini C18 column (250 × 3.0 mm, 5 μm) preceded by a
209 security guard column C18 (4 × 3 mm, 5 μm), (Phenomenex, Torrance, CA, USA). The volume of
210 injection was 20 μL, and the wavelength of detection was 254 nm.

211 The analytical method was validated (Table S1) showing a good linearity ($R^2 \geq 0.995$) for all the 4
212 detected phytosterols at a concentration range from 0.5 to 100 μg/mL. Moreover, the reproducibility
213 of the method was determined calculating the relative standard deviation (%RSD) between 3
214 consecutive analysis of a mix standard sample (50 μg/mL) on the same day (intraday) and during 3
215 consecutive days (interday). The intraday reproducibility was between 0.1% and 1.8% while the
216 interday reproducibility ranged between 1.2 – 6.3% confirming thus the reproducibility of the
217 method. The assessment of the LOD (9-25 ng/mL) and the LOQ (29-83 ng/mL) showed the high
218 sensibility of the analytical method.

219 2.76-Statistical analyses

220 All the experiments were carried out in three replicates ($n=3$) and the results are expressed as
221 mean ± standard deviation (SD). Relative standard deviation (%RSD) was determined between
222 replicates to assess the precision of the obtained results by using Microsoft Excel (Microsoft Office
223 2019).

224 3. Results and discussions

225 3.1 Macronutrients and fiber composition of CSS Nutritional composition

226 The nutritional composition (moisture, carbohydrates, fat, protein, dietary fiber, and ash content) of
227 CSS is reported in Table 32. CSS is mainly composed of dietary fibers (51.5 ± 9.7 g/100 g), which
228 are important components for functional foods dedicated to promoting the digestive system health
229 and reduce the risk of chronic diseases such as cardiovascular diseases and type-2 diabetes (Yegin,
230 Kopec, Kitts, and Zawistowski, 2020). Moreover, CSS resulted to be high in proteins with a total

231 protein content of 14.2 ± 1.0 g/100g. CSS could thus represent a novel source of dietary proteins to
232 address the global protein challenge associated with the growing population (Weindl et al., 2020).
233 In addition, CSS shows relatively low levels of fats (3.7 ± 0.3 g/100 g) and consistent ash content
234 (4.9 %). Furthermore, CSS shows also a low moisture content (7.70 ± 0.46 %), which is reported to
235 facilitate the grinding process and allow the production of small particle sizes with reduced energy
236 (Moon & Yoon, 2018). These characteristics are particularly advantageous for possible industrial
237 transformations of CSS.

238 The nutritional composition of CSS is consistent with the range of levels reported in the literature.
239 Indeed, Wen et al (2020) reported 14.6 g/100 g of proteins and fat content of 3.6 g/100 g in CSS,
240 while Ballesteros, Teixeira, and Mussatto (2014) obtained dietary fiber levels of 54.1 g/100 g.

241 **3.2 Mineral composition of CSS**

242 The analysis of minerals was also performed for a total of 30 targeted minerals including 7
243 macrominerals (calcium, chloride, magnesium, phosphorus, potassium, sodium, and sulfur). To our
244 knowledge, the present study reports the most complete mineral composition of CSS with a total of
245 30 quantified minerals (Table 32).

246 Analyses highlighted that CSS possesses high levels of calcium (1,080 mg/100 g) and can be
247 exploited for the production of calcium ingredients for food or pharmaceutical applications. Indeed,
248 calcium is an essential element for various physiological functions including neurotransmission,
249 blood clotting, cell division, muscle activity, and bone and teeth maintenance (Cormick et al.,
250 2021). Moreover, CSS showed high levels of potassium (972 mg/100 g). This macromineral plays a
251 key role in the maintenance of blood pressure, the nervous system, and muscle functions (Udensi &
252 Tchounwou, 2017). In addition, CSS resulted to be a good source of magnesium being the third
253 most abundant mineral of CSS (257 mg/100 g). Magnesium is an essential mineral reported to play
254 numerous roles in the normal functioning of the human body such as energy production, muscle
255 contraction, protein synthesis, bone growth, and blood pressure (Al Alawi, Majoni, & Falhammar,
256 2018). Appreciable levels of chloride (12.3 mg/kg), phosphorus (12.4 mg/kg), sodium (110 mg/kg),

257 and sulfur (51.9 mg/kg) were also observed. Although some differences related to the studied coffee
258 varieties, calcium, potassium, and magnesium are reported in similar studies to be the most
259 abundant minerals in CSS. [Ballesteros et al. \(2014\)](#) reported higher levels of potassium (2.11 g/100
260 g) in CSS, which was more abundant than calcium (0.94 g/100 g) and magnesium (0.31 g/100 g).
261 Contrarily, [Wen et al. \(2020\)](#) obtained a more similar order with calcium, potassium, and
262 magnesium levels of 1.48, 1.21, and 0.05 g/100 g respectively.

263 Concerning microminerals, 23 elements were assessed with a total concentration of 540.4 mg/kg in
264 CSS. Iron resulted to be the most concentrated micromineral in CSS with levels of 238 mg/kg
265 followed by aluminum (89.0 mg/kg). Iron is an important micronutrient playing key roles in red
266 blood cell formation, cell division, fatigue reduction, immune system, and cognitive functions
267 ([Finkelstein, Haas, & Mehta, 2017](#)). In addition, high levels of essential micronutrients were also
268 observed including manganese (46.7 mg/kg), copper (37.9 mg/kg), zinc (31.9 mg/kg), boron (26.1
269 mg/kg), and molybdenum (0.2 mg/kg). These results are consistent with the study of [Ballesteros et](#)
270 [al. \(2014\)](#). Copper and manganese contribute to bone health maintenance, oxidative stress
271 modulation, and connective tissue formation. In addition, copper supports the nervous and immune
272 systems and is essential for the absorption and transport of iron ([Cámara et al., 2021](#)). Boron is
273 reported to be involved in various functions such as bone growth and maintenance, magnesium
274 absorption, and oxidative stress reduction ([Pizzorno, 2015](#)). Zinc is an essential micronutrient
275 supporting a wide number of functions including metabolism, bone health, immune system, cell
276 division, protein synthesis, cognitive function, reproduction, and cell protection from oxidative
277 stress ([Cakmak & Kutman, 2018](#)).

278 However, significant levels of potentially toxic and carcinogenic heavy metals were also observed
279 including arsenic (0.13 mg/kg), cadmium (0.07 mg/kg), lead (0.25 mg/kg), or mercury (0.05
280 mg/kg), which can cause various health hazards. Therefore, it is necessary to apply decontamination
281 protocols using chelating agents for the removal of heavy metals before the valorization of CSS in
282 food or nutraceutical products.

3.3 Content of Vitamin B₂ and B₃ in CSSB-vitamins analysis in CSS

3.3.1 Analytical method validation

The analysis of vitamin B₂ (riboflavin) and B₃ (nicotinic acid and nicotinamide) has been carried out following a developed method (Caprioli et al., 2018) which used a hydrophilic interaction chromatography (HILIC) instead of classic reversed phase (RP) since HILIC is more suitable for polar compounds as hydrosoluble vitamins are, and it enhances compatibility with MS system (Porter, & Lodge, 2021). Before the analysis of CSS, the analytical procedure has been validated by assessing the linearity, sensitivity, repeatability, and recovery (Table 3). The linearity was evaluated by injecting eight different concentrations of standard mixtures and preparing calibration curves. These were constructed by plotting the standard peak areas by concentrations and for each analyte, a determination coefficient (R^2) was obtained. All analytes showed good linearity since the R^2 was ≥ 0.9945 . The sensitivity has been studied by measuring the limit of detection (LOD) and limit of quantification (LOQ). LOD and LOQ were estimated by injecting low concentrations of studied compounds and the concentration that generated a signal-to-noise ratio (SNR) of 3 was assigned to LOD while that with SNR of 10 was designated as LOQ. The analytes showed satisfactory sensitivity since LOQ was 3 ng/mL for all vitamins and these values were similar to those reported in the literature (Caprioli et al., 2018). The method repeatability was evaluated by injecting five times three analyte concentrations in the same day and during three consecutive days and the intraday and interday repeatability was calculated and expressed as relative standard deviation (%RSD). The intraday and interday repeatability were 2.5–4.6% and 3.9–6.2%, respectively. Recovery studies were performed at two different spiking levels, 1 and 0.1 $\mu\text{g/g}$ to study the accuracy of the current method. Satisfactory recovery levels were found with both levels for all analytes such as 78–86% with 0.1 $\mu\text{g/g}$ and 82–93% with 1 $\mu\text{g/g}$.

3.3.2 Content of B-vitamins

This is the first study that investigated the content of vitamin B₂ (riboflavin) and B₃ (niacin) in CSS. Table 4 reports the content, expressed as $\mu\text{g/g}$, of riboflavin (vitamin B₂), nicotinic acid,

309 nicotinamide, and niacin (vitamin B₃) found in coffee silverskin. As an example, **Figure 1** reports
310 the HPLC-MS/MS chromatograms of a standard mixture at 100 ng/mL and a CSS sample. Three
311 different SSR such as 20:1, 30:1, and 40:1 (v/w, mL/g) were tested in the current work. Several
312 papers reported that varying SSR can influence the extraction of bioactive compounds (Papoutsis et
313 al., 2018). For this reason, in the present research, the extraction has been carried out by varying the
314 SSR and tiny differences have been noticed in the vitamin content extracted using diverse SSR. In
315 detail, SSR of 30:1 and 20:1 seemed to slightly increase the extraction of vitamins.
316 Niacin (2.51–3.07 µg/g), i.e., nicotinic acid (1.88–2.33 µg/g) and nicotinamide (0.63–0.74 µg/g),
317 occurred at higher levels than riboflavin (0.18–0.20 µg/g) in CSS. These results are consistent with
318 those reported in the literature in which niacin occurred at a higher level than riboflavin in coffee
319 beverages (Wachamo, 2017). The levels of niacin especially that of nicotinic acid varies according
320 to the brewing method and roasting profile since it is derived from the second main alkaloids in
321 coffee, trigonelline, which during roasting is degraded into several volatile (pyridine and pyrrole
322 derivatives) and non-volatile compounds (nicotinic acid) (Jeszka-Skowron, Frankowski, & Zgoła-
323 Grzeškowiak, 2020). The presence of these two important vitamins, which act as co-factors in
324 numerous essential enzymatic reactions fundamental to maintain the body homeostasis (Thakur et
325 al., 2017), highlights the coffee silverskin value and leads the way for innovative reuses.

326 **3.4 Phytosterol composition of CSS**

327 The analysis of phytosterols (PS) was performed in CSS using a developed and validated
328 derivatization method allowing the analysis of main sterols in HPLC at higher and more selective
329 detection wavelengths. Few studies assessed the content of PS in CSS. Analyses allowed the
330 identification and quantification of 4 sterols: Δ^5 -avenasterol, β -sitosterol, campesterol, and
331 stigmasterol for a total concentration of 98.4 ± 8.0 mg/kg (**Table 5**). β -sitosterol was the most
332 abundant sterol (77.1 ± 2.8 mg/kg) followed by campesterol (10.9 ± 3.1 mg/kg), stigmasterol ($9.9 \pm$
333 2.2 mg/kg), and Δ^5 -avenasterol (0.5 ± 0.1 mg/kg). These results are concordant with the study of
334 Toschi et al (2014). CSS can be considered as a potential source of PS, which are highly valorized

335 in food and nutraceutical industries due to their hypocholesterolemic effect. Del Pozo et al. (2021)
336 proposed the conversion through pyrolysis of CSS into value-added products. The obtained
337 pyrolysis liquids were rich in caffeine, polyphenols, and β -sitosterol. CSS could thus be
338 transformed into a PS-enriched ingredient with high value-added applications. However, the high
339 temperatures reached during coffee roasting (170–260 °C), lead to the oxidation of PS in CSS.
340 Indeed, PS are susceptible to oxidation at high temperatures forming phytosterol oxidation products
341 (POPs) including 7-hydroxysterols, 7-ketosterols, 5,6-epoxysterols, and 3,5,6-triols, which are
342 reported to possess pro-atherogenic and pro-inflammatory properties (Kamgang Nzekoue et al.,
343 2020). In CSS, Toschi et al (2014) reported considerable levels of POPs with an oxidation rate
344 between 27.6 % and 48.1 %. Therefore, the valorization of CSS for PS production will necessarily
345 require the prior removal of POPs using separation techniques such as solid-phase extraction
346 (Azadmard-Damirchi & Dutta, 2009).

347 **3.5.4. Discussion**

348 Consistent amounts of by-products are generated during the coffee industrial processing and
349 according to FAO, growth projections on coffee production and consumption are expected to
350 increase over the next years. In 2019 the world production accounted for 10,035,576 tonnes with a
351 harvested area of 11,120,498 hectares, and data compared to 2009 respectively account for + 1.5%
352 in production and + 6.31% in the harvesting surface (FAO, 2019). Wastes generated from the coffee
353 industries should be considered useful resource and some of them have already become a resource,
354 especially in agricultural contexts. For example, spent coffee grounds are used with a «Zero
355 Wastes» approach with different applications in agriculture, such as fertilizer for growing
356 mushrooms, as they contain minerals and nutrients that are useful for their growth. Other
357 applications contemplate the use of coffee wastes to create pellet or to produce biobased materials
358 such as pens and cups.

359 Consideration should be given to the environmental impacts caused by these agricultural by-
360 products, especially in regions dedicated to its cultivation and processing, such as in Latin America

361 (Iriundo-DeHond, et al., 2017). Moreover, in developing countries, such as Uganda, the importance
362 of the coffee sector is relevant as a key driver of rural economic activity and income sources
363 (Mwesigye & Nguyen, 2020). Therefore, a sustainable and fair coffee value chain is a key element
364 in the valorization of coffee wastes in a Circular Economy approach in order to cope with
365 environmental issues and market instability.

366 CSS presents the advantage to be easily collected in high quantity due to its production in coffee-
367 roasting industries. Moreover, CSS is a more stable material, with a low moisture content (ranging
368 between 5 and 10%) and a lower microbial charge due to the high temperatures reached during
369 roasting (Bessada et al., 2018). Compared to other coffee by-products like spent coffee ground, CSS
370 is thus easily exploitable and more advantageous for industrial transformations.

371 The present research highlighted significant levels of healthy nutrients and bioactive compounds in
372 CSS (proteins, minerals, dietary fibers, ...), which can thus be used in the development of
373 functional ingredients for food and pharmaceutical uses. Attention should be given to the
374 antinutritional factors and toxic compounds revealed upper (POPs, heavy metals), which have to be
375 removed before potential valorization as food ingredients.

376 Moreover, this study assessed and reported for the first time the levels of B-vitamins (B₂ and B₃) in
377 CSS confirming the initial hypothesis. The increase of knowledge on CSS will speed up the design
378 of novel value-added applications for CSS. The adoption of certified sustainable production of
379 coffee is estimated to bring positive social impacts while implying increased farm incomes with
380 more stable employment possibilities for farmworkers, thus enhancing community development as
381 well as reducing environmental and health risks caused by the accumulation of coffee wastes (Word
382 Bank, 2019).

383 **4.5. Conclusions**

384 The growing research of value-added applications for CSS valorization is nowadays a priority
385 research line to face the environmental challenges of the coffee production system and by-products
386 accumulation. CSS valorization requires analytical studies, in order to quantify healthy nutrients

387 and bioactive compounds, which could be extracted to develop functional ingredients with potential
388 applications in the food, pharmaceutical, or cosmetic sectors. In this study, the profile of the less
389 studied compounds in CSS including microminerals, phytosterols, and vitamins B₂ and B₃ was
390 assessed and reported. The novelty accompanying the obtained results, confirms the under-
391 exploration of CSS potentialities. Further studies are thus required for a comprehensive evaluation
392 of CSS properties to increase the sustainability of the coffee agro-industry.

393 **Aknologements**

394 This work was carried out within the Framework Agreement between illycaffè spa and the
395 University of Camerino signed on August 6, 2021.

396

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516 Highlights

- 517 1- Nutritional composition of coffee silverskin (CSS) from arabica coffee.
- 518 2- Comprehensive assessment of macrominerals (7) and microminerals (23) in CSS.
- 519 3- Vitamin B₃ (2.51–3.07 µg/g) and vitamin B₂ (0.18–0.20 µg/g) are present in CSS.
- 520 4- Phytosterols analysis revealed high levels of β-sitosterol (77.1 ± 2.8 mg/kg).

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522 **Highlights**

Compound	Retention time (Rt) (min)	Precursor ion (<i>m/z</i>)	Product ion (<i>m/z</i>)	Fragmentor (V)	Collision energy (V)	Polarity
Nicotinic acid	1.6	124	80 ^a	92	21	positive
			53		33	
Nicotinamide	1.7	123	80 ^a	92	21	positive
			53		33	
Riboflavin	2.1	377	243 ^a	131	21	positive
			172		41	

523 5- Nutritional composition of coffee silverskin (CSS) from arabica coffee.

524 6- Comprehensive assessment of macrominerals (7) and microminerals (23) in CSS.

525 7- Vitamin B₃ (2.51–3.07 µg/g) and vitamin B₂ (0.18–0.20 µg/g) are present in CSS.

526 8- Phytosterols analysis revealed high levels of β-sitosterol (77.1 ± 2.8 mg/kg).

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528 **Table 1.** HPLC-MS/MS acquisition parameters used for the quantification of niacin (nicotinic acid
529 and nicotinamide) and riboflavin

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533 ^a These product ions were used for quantification, the other for confirming the analytes.

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537 **Table 1.** HPLC-MS/MS acquisition parameters used for the quantification of niacin (nicotinic acid
 538 and nicotinamide)
 540 and riboflavin

Compounds	Unit	Levels ^a
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544 ^a These product ions were used for quantification, the other for confirming the analytes.

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Compound	Concentration range (ng/mL)	Regression equation	R ²	LOD (ng/mL)	LOQ (ng/mL)	Intraday repeatability (RSD%)	Interday repeatability (RSD%)
Nicotinic acid	3–5000	$y = 17500x + 1154.1$	0.9971	1	3	2.5	4.2
Nicotinamide	3–5000	$y = 25894x + 2369.6$	0.9953	1	3	2.7	3.9
Riboflavin	3–5000	$y = 11855x + 1163$	0.9945	1	3	4.6	6.2

548 **Table 23.** HPLC-MS/MS validation parameters such as linearity, sensitivity, repeatability, and
 549 recovery

550 44.

Compound	Concentration range (ng/mL)	Regression equation	R ²	LOD (ng/mL)	LOQ (ng/mL)	Intraday repeatability (RSD%)	Interday repeatability (RSD%)
Nicotinic acid	3–5000	$y = 17500x + 1154.1$	0.9971	1	3	2.5	4.2
Nicotinamide	3–5000	$y = 25894x + 2369.6$	0.9953	1	3	2.7	3.9
Riboflavin	3–5000	$y = 11855x + 1163$	0.9945	1	3	4.6	6.2

551 **Table 24.** HPLC-MS/MS validation parameters such as linearity, sensitivity, repeatability, and
 552 recovery

553 45.

554 **Table 32.** Nutritional composition and minerals profile of coffee silverskin

Nutrients

Ash	g/100 g	4.9 ± 0.1
Carbohydrates	g/100 g	18.0 ± 3.0
Fats	g/100 g	3.7 ± 0.3
Moisture	g/100 g	7.7 ± 0.5
Proteins	g/100 g	14.2 ± 1.0
Total dietary fibers	g/100 g	51.5 ± 9.7

Macrominerals

Calcium	mg/100 g	1080 ± 69
Chloride	mg/kg	12.3 ± 0.4
Magnesium	mg/100 g	257 ± 18
Phosphorus	mg/100 g	12.4 ± 0.8
Potassium	mg/100 g	972 ± 46
Sodium	mg/100 g	11 ± 1
Sulphur	mg/kg	51.9 ± 1.3

Microminerals

Aluminum	mg/kg	89.0 ± 6.5
Antimony	mg/kg	0.05 ± 0.009
Arsenic	mg/kg	0.1 ± 0.04
Barium	mg/kg	66.75 ± 12.4
Beryllium	mg/kg	≤0.01
Boron	mg/kg	26.1 ± 4.8
Cadmium	mg/kg	0.07 ± 0.01
Chrome	mg/kg	0.23 ± 0.02
Cobalt	mg/kg	0.2 ± 0.06
Copper	mg/kg	37.9 ± 9.7
Iron	mg/kg	238 ± 11
Lead	mg/kg	0.3 ± 0.005
Manganese	mg/kg	46.7 ± 2.8
Mercury	mg/kg	0.05 ± 0.002
Molybdenum	mg/kg	0.2 ± 0.04
Nickel	mg/kg	0.5 ± 0.01

555		Selenium	mg/kg	0.1 ± 0.03	
556		Silver	mg/kg	0.03 ± 0.006	
557		Compounds	Unit	Levels	
558		Nutrients	mg/kg	0.05 ± 0.004	
559		Ash	g/100 g	4.9 ± 0.1	
560		Carbohydrates	g/100 g	18.0 ± 3.0	
562	a All the experiments were carried out in three replicates (n=3) and the results are expressed as mean ± standard deviation (SD).	Fats	g/100 g	3.7 ± 0.3	
563		Moisture	g/100 g	7.7 ± 0.5	
566		Proteins	g/100 g	14.2 ± 1.0	
568		Total dietary fibers	g/100 g	51.5 ± 9.7	
569					
570		46.			
571			Macrominerals		
572	Table 32. composition profile of silverskin	Calcium	mg/100 g	1080 ± 69	Nutritional and minerals coffee
573		Chloride	mg/kg	12.3 ± 0.4	
575		Magnesium	mg/100 g	257 ± 18	
577		Phosphorus	mg/100 g	12.4 ± 0.8	
		Potassium	mg/100 g	972 ± 46	
		Sodium	mg/100 g	11 ± 1	
		Sulphur	mg/kg	51.9 ± 1.3	
		Microminerals			
		Aluminum	mg/kg	89.0 ± 6.5	
		Antimony	mg/kg	0.05 ± 0.009	
		Arsenic	mg/kg	0.1 ± 0.04	
		Barium	mg/kg	66.75 ± 12.4	

578		Beryllium	mg/kg	≤ 0.01	
579		Boron	mg/kg	26.1 ± 4.8	
580		Cadmium	mg/kg	0.07 ± 0.01	
581		Chrome	mg/kg	0.23 ± 0.02	
582		Cobalt	mg/kg	0.2 ± 0.06	
583		Copper	mg/kg	37.9 ± 9.7	
584	<p>a All the experiments were carried out in three replicates ($n=3$) and the results are expressed as mean \pm standard deviation (SD).</p> <p>47.</p> <p>Table 4.</p> <p>of niacin and coffee using different sample ratios 20:1, 30:1 and</p>	Iron	mg/kg	238 ± 11	experiments in three
586		Lead	mg/kg	0.3 ± 0.005	and the results
588		Manganese	mg/kg	46.7 ± 2.8	mean \pm standard
590		Mercury	mg/kg	0.05 ± 0.002	
592		Molybdenum	mg/kg	0.2 ± 0.04	
593		Nickel	mg/kg	0.5 ± 0.01	Content ($\mu\text{g/g}$)
594		Selenium	mg/kg	0.1 ± 0.03	riboflavin in
596		Silver	mg/kg	0.03 ± 0.006	silverskin
598		Thallium	mg/kg	≤ 0.01	
600		Tin	mg/kg	0.05 ± 0.004	solvent-to-
602	Titanium	mg/kg	2.1 ± 0.1	(SSR), i.e.,	
604	Vanadium	mg/kg	0.2 ± 0.06	40:1	
606	Zinc	mg/kg	31.9 ± 5.3		

Compounds	SSR ^a (mL/g)		
	20:1	30:1	40:1
Niacin (B ₃)	2.92 ± 0.10	3.07 ± 0.13	2.51 ± 0.16
- Nicotinamide	0.70 ± 0.06	0.74 ± 0.06	0.63 ± 0.02
- Nicotinic acid	2.22 ± 0.03	2.33 ± 0.07	1.88 ± 0.13
Riboflavin (B ₂)	0.18 ± 0.01	0.20 ± 0.01	0.18 ± 0.01

607 ^a All the experiments were carried out in three replicates ($n=3$) and the results are expressed as
 608 mean \pm standard deviation (SD).

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612 **Table 4.**613 ($\mu\text{g/g}$) of

615 riboflavin

618 silverskin

620 different

622 sample

624 (SSR), i.e.,

625 and 40:1

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Compounds	Unit	SSR ^a (mL/g)			Content niacin and in coffee using solvent-to- ratios 20:1, 30:1 and 40:1
		20:1	30:1	40:1	
Campesterol	mg/kg				10.9 ± 2.1
Stigmasterol	mg/kg				9.9 ± 1.6
β -Sitosterol	mg/kg				77.1 ± 2.8
Δ 5-Avenasterol	mg/kg				0.5 ± 0.1
Niacin (B ₃)		2.92 ± 0.10	3.07 ± 0.13	2.51 ± 0.16	
- Nicotinamide		0.70 ± 0.06	0.74 ± 0.06	0.63 ± 0.02	
- Nicotinic acid		2.22 ± 0.03	2.33 ± 0.07	1.88 ± 0.13	
Riboflavin (B ₂)		0.18 ± 0.01	0.20 ± 0.01	0.18 ± 0.01	

627 ^a All the experiments were carried out in three replicates ($n=3$) and the results are expressed as
 628 mean \pm standard deviation (SD).

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631 **Table 5.** Concentrations of phytosterols in coffee silverskin

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Compounds	Unit	Levels ^a
Campesterol	mg/kg	10.9 ± 2.1
Stigmasterol	mg/kg	9.9 ± 1.6
β-Sitosterol	mg/kg	77.1 ± 2.8
Δ5-Avenasterol	mg/kg	0.5 ± 0.1

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mean ± standard deviation (SD).

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Table 5. Concentrations of phytosterols in coffee silverskin

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^a All the experiments were carried out in three replicates ($n=3$) and the results are expressed as mean ± standard deviation (SD).

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