

## Article

# Characterization of Polyphenols and Volatile Compounds from Understudied Algerian *Pallenis spinosa* by HS-SPME-GC-MS, NMR and HPLC-MS<sup>n</sup> Approaches

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**Abstract:** *Pallenis spinosa* (L.) Cass. is a widespread plant in the Mediterranean region. Traditionally, it is used as a medicinal species to treat several ailments, from inflammation to skin injuries. Although the phytochemical content of this plant has already been investigated, there is currently limited data on Algerian *P. spinosa*. In this work, we focused on volatile compounds and non-volatile secondary metabolites extracted using HS-SPME and methanol from the aerial parts of *P. spinosa* collected from Northeast Algeria. Volatile constituents were analyzed by GC-MS, while non-volatile compounds were analyzed by NMR and HPLC-MS<sup>n</sup>. In total, 48 volatile compounds were identified, including sesquiterpene hydrocarbons (65.8%), monoterpene hydrocarbons (16.9%), and oxygenated monoterpenes and sesquiterpenes (8.3% and 6.5%, respectively).  $\beta$ -Chamigrene (16.2%),  $\alpha$ -selinene (12.8%),  $\beta$ -pinene (10.6%), and  $\beta$ -caryophyllene (9.2%) were assessed as the main constituents. Concerning non-volatile metabolites, 23 polyphenols were identified (7.26 mg/g DW), and phenolic acids were predominant (5.83 mg/g DW). Tricaffeoylhexaric acid (1.76 mg/g DW), tetracaffeoylhexaric acid (1.41 mg/g DW), 3,5-dicaffeoylquinic acid (1.04 mg/g DW), caffeoyl dihexoside (0.35 mg/g DW), and chlorogenic acid (0.29 mg/g DW) were the most abundant ones. Several known flavonoids, such as tricetin and patuletin glycosides, kaempferol, and apigenin, were also identified, and myricetin hexoside was detected in *P. spinosa* for the first time. Overall, our work is the first to report an exhaustive characterization of volatile and non-volatile secondary metabolites from Algerian *P. spinosa*. The results represent a step forward in revealing the chemistry of this widespread plant species. Furthermore, they may contribute to rationalizing its traditional medicinal applications and preserve the biodiversity of Algerian flora.

**Keywords:** *Pallenis spinosa*; secondary metabolites; terpenes; polyphenols; headspace analysis; mass spectrometry



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## 1. Introduction

The family of Asteraceae contains between 25,000 and 30,000 species belonging to over 1000 genera. *Pallenis spinosa* (L.) Cass., commonly known as the spiny starwort, is a member

of this family, belonging to the tribe of Inuleae, a subtribe of Inulineae, and a group of *Inula* [1]. The genus *Pallenis* is monotypic and is widespread in southern Europe, northern Africa, the Canary Islands, the Middle East, and the Mediterranean region, especially in the desert and in the coastal habitats [2]. It is an annual herbaceous plant characterized by large yellow/orange flowers surrounded by long spiny involucre bracts [3]. Traditionally, the flowers and the spiny leaves of the plant have been used as medicine to treat inflammation, gastralgia, mouth infections, skin injuries, and inflammatory contusions [4,5]. Usually, the flowers and the leaves are taken orally or applied topically in the form of infusion or decoction [6].

In the western part of Algeria, *P. spinosa* is used for the management of skin conditions such as eczema. Furthermore, its flowers hold significant medicinal value in addressing sensitivity issues, as well as aiding in the healing of bruises and wounds. Across the Mediterranean basin, the plant is used in its entirety as an insecticide [7,8].

Some studies have investigated the secondary metabolic products of *P. spinosa* in order to improve traditional knowledge and find novel medicinal applications. Regarding volatile constituents, the plant is rich in sesquiterpenes and oxygenated sesquiterpenoids, including a ketone with a previously unreported carbon skeleton (dihydroxypallenone) [9,10]. Other compounds such as  $\alpha$ -cadinol, shiromool, germacrene D derivatives [11], and germacrene A-type epoxides [12] have been reported. Concerning non-volatile secondary metabolites, *P. spinosa* tends to accumulate 5-O-glycosyl flavones. Among these, several tricetin and patuletin derivatives have been detected in their aerial parts [13].

Natural products have provided many drugs and drug leads and still remain the richest sources of bioactive compounds. In this work, we aimed at moving a step forward in the characterization of secondary metabolites from *P. spinosa*. Specifically, we investigated volatile compounds and secondary metabolites extracted from the aerial parts of *P. spinosa* from Northeastern Algeria (Bordj Ghedir). Although the phytochemical content of this plant has already been studied, limited data exists on the species growing in Algeria. Headspace solid-phase micro-extraction coupled with gas chromatography-mass spectrometry (HS-SPME-GC-MS), nuclear magnetic resonance (NMR), and high-performance liquid chromatography-mass spectrometry (HPLC-MS<sup>n</sup>) were used for the analyses, respectively. Overall, the results are intended to contribute to the valorization of understudied Algerian medicinal and edible plants and the rationalization of their traditional uses.

## 2. Materials and Methods

### 2.1. Chemicals and Reagents

HPLC-grade methanol and acetonitrile, n-hexane, dichloromethane, ethyl acetate, butanol, and formic acid were purchased from Sigma-Aldrich (Milan, Italy). Ultrapure water was used, and it was obtained using a Milli-Q<sup>®</sup> dispenser (Merck, Darmstadt, Germany). Rutin and chlorogenic acid reference standards were purchased from Sigma-Aldrich (Milan, Italy). Individual standard solutions of rutin and chlorogenic acid were prepared at 1 mg/mL in methanol and were stored at  $-20\text{ }^{\circ}\text{C}$  until use. Sequential dilutions of the stock solutions were prepared in methanol, and the obtained samples were used to build calibration curves, as described in the following paragraphs.

### 2.2. Plant Material

Aerial parts of *P. spinosa* were harvested at the full flowering period during the month of May 2020 from the Bordj Ghedir Region, Province of Bordj Bou Arreridj, North-East Algeria, at an altitude of 820 m a.s.l. Botanical identification of the plant was carried out based on morphological features with the help of authentic flora [5], and a voucher specimen was deposited in the Herbarium of M'vila University (MUNIV001220). Before extraction, plant material was washed with tap water to remove dust and dried in the shade in a well-ventilated place for 7 days. A part of the material was used fresh for the extraction of volatiles.

### 2.3. HS-SPME-GC-MS Analysis of Volatile Compounds

For HS-SPME, the method described by Ascrizzi et al. [14] was applied. Briefly, the fresh plant material (2 g) was placed in a 4 mL glass vial, which was sealed and left to equilibrate at room temperature for 1 h. Afterwards, the headspace was sampled with a Supelco SPME fiber coated with polydimethylsiloxane (PDMS, 100  $\mu\text{m}$ ) for 30 min. GC-MS analysis of the headspace extract was performed as described by Bendif et al. [15], using a Varian CP-3800 gas chromatograph coupled to a Varian Saturn 2000 mass spectrometer. An Agilent DB-5 capillary column (30 m  $\times$  0.25 mm; coating thickness 0.25  $\mu\text{m}$ ) was used as the stationary phase. Chromatographic conditions were as follows: injector and transfer line temperatures, 220 and 240  $^{\circ}\text{C}$ , respectively; oven temperature programmed from 60 to 240  $^{\circ}\text{C}$  at 3  $^{\circ}\text{C}/\text{min}$ ; helium was used as carrier gas at 1 mL/min; splitless injection. Identification of volatile compounds was performed by comparing retention times with those of reference pure compounds, comparing their linear retention indices (LRI) relative to the  $\text{C}_6$ – $\text{C}_{28}$  series of n-hydrocarbons, and by matching their mass spectra against commercial [16,17] and home-made libraries built up from pure substances.

### 2.4. Preparation of Extracts for NMR and HPLC-MS<sup>n</sup> Analyses

Sample preparation was performed using a protocol previously described by Dall'Acqua et al. [18] and Bendif et al. [15]. A total of 20 g of the dried aerial parts were ground with an IKA A11 basic analytical mill to obtain a fine powder. The powder was suspended in 150 mL of methanol and then sonicated for 10 min. After centrifugation, the supernatant was recovered in an Erlenmeyer flask, and the residue was re-extracted with a further 50 mL of the same solvent two more times. Thereafter, supernatants were collected and concentrated under reduced pressure with a rotary evaporator at 35  $^{\circ}\text{C}$ , yielding 2.13 g of crude extract (yield 10.65%, *w/w*). An amount of the dry extract was stored in dark glass vials at  $-20^{\circ}\text{C}$  until its chemical characterization. Two grams were used for fractionation using solvents at increasing polarity. The extract was suspended in a methanol/water (1:9) mixture (50 mL) and sonicated. The mixture was then partitioned with n-hexane (HEX; 20 mL, 3 times), dichloromethane (DCM; 20 mL, 3 times), ethyl acetate (EA; 20 mL, 3 times), and butanol (BUT; 20 mL, 3 times). The fractions were dried under vacuum and analyzed by NMR.

### 2.5. NMR Analysis

The crude extract and dried fractions were suspended in deuterated methanol.  $^1\text{H}$ -NMR spectra were recorded with a BRUKER ADVANCE III spectrometer (Bruker, Billerica, MA, USA) operating at 400 MHz, using standard pulse sequences.

### 2.6. HPLC-MS<sup>n</sup> Analysis

The HPLC-MS<sup>n</sup> method was used for the identification of secondary metabolites in the crude methanolic extract of *P. spinosa*. Before analysis, the dry sample was dissolved in methanol at a concentration of 5 mg/mL and filtered through a 0.45  $\mu\text{m}$  Millipore filter. The HPLC system was composed of a Varian 212 binary pump equipped with a Varian Prostar 430 autosampler coupled to a Varian 500 Ion Trap mass detector (MS). An Agilent Eclipse plus C18 column (2.1  $\times$  150 mm, 3.5  $\mu\text{m}$ ) was used as the stationary phase. A gradient of acetonitrile (A) and 0.1% *v/v* formic acid in water (B) was used as the mobile phase. The gradient was set as follows: 0 min, 10% A; 20 min, 54% A; 23 min, 100% A; isocratic up to 32 min. Re-equilibration time was 8 min. The flow rate was 0.2 mL/min. MS data were acquired in the negative ion mode [ESI (-)], and the operating parameters of the spectrometer were as follows: needle voltage, 4500 V; capillary voltage, 70 V; RF loading, 100%; nebulizing gas pressure, 20 psi (nitrogen); dry in gas pressure, 15 psi; dry in gas temperature, 350  $^{\circ}\text{C}$ . Mass spectra were recorded in the range of *m/z* 50–2000. The turbo detection data scanning (TDDS) function of the instrument was used to investigate the fragmentation pattern of the eluted compounds, setting *n* = 3 levels of fragmentation. Fragmentation data, together with information on the molecular weight of the compounds,

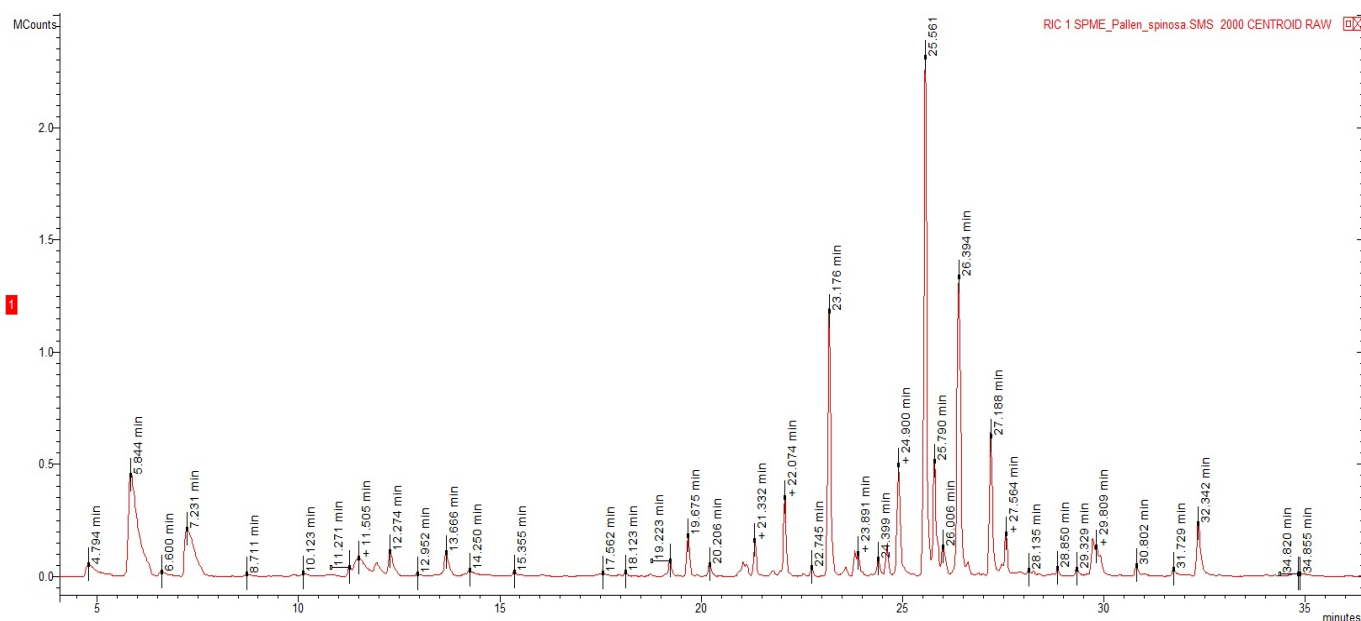
were used to perform the tentative identification of secondary metabolites from *P. spinosa*. As a comparison, literature data were evaluated.

Identified phenolic compounds were quantified using linear calibration curves. These latter were calculated by analyzing standard solutions of rutin (for flavonoids) and chlorogenic acid (for phenolic acids) and integrating the corresponding peaks to obtain the area under the curve (AUC) values. Concentration ranges were 12–120 µg/mL and 10.4–104 µg/mL for rutin and chlorogenic acid, respectively. The equations were  $y = 7342x - 19,707$  ( $R^2 = 0.9999$ ) for rutin, and  $y = 7504.8x - 3890.2$  ( $R^2 = 1$ ) for chlorogenic acid. The analysis was performed in triplicate, and the results were expressed as mean  $\pm$  standard deviation (S.D.).

### 3. Results and Discussion

#### 3.1. HS-SPME-GC-MS Analysis of Volatile Compounds

The development of quick and sustainable methods for the extraction of volatile compounds from plant materials such as the HS-SPME is important. The rapid sampling shortens the whole length of the analysis, while the avoidance of organic solvents reduces its environmental impact. HS-SPME coupled with the GC-MS analysis of the aerial parts of *P. spinosa* exhibited 48 volatile compounds, accounting for 98.1% of the whole volatile extract. The identified compounds, their retention indexes, and their relative proportions are presented in Table 1. A representative chromatogram is shown in Figure 1.



**Figure 1.** Exemplificative chromatogram obtained from the GC-MS analysis of the volatile extract obtained from *P. spinosa* by HS-SPME.

The volatile profile was dominated by sesquiterpene hydrocarbons (65.8%), followed by monoterpene hydrocarbons (16.9%). Oxygenated monoterpenes were less represented, similar to oxygenated sesquiterpenes (8.3% and 6.5%, respectively). Phenylpropanoids (0.5%) and apocarotenes (0.1%) were found as trace components. Considering the single identified compounds, the aerial parts of *P. spinosa* were characterized mainly by  $\beta$ -chamigrene (16.2%),  $\alpha$ -selinene (12.8%),  $\beta$ -pinene (10.6%), and  $\beta$ -caryophyllene (9.2%). Some other compounds, such as trans- $\gamma$ -cadinene (5.2%), alloaromadendrene (4.4%), and germacrene D (4.3%), were found.

**Table 1.** Aroma profile (relative percentages of identified compounds) of the aerial parts of *P. spinosa* obtained using HS-SPME–GC–MS. Compounds are listed in order of their elution from an HP-5MS column.

Constituents	Class <sup>a</sup>	LRI <sup>b</sup>	RI Lit. <sup>c</sup>	%
α-pinene	MH	941	939	1.0
β-pinene	MH	982	979	10.6
α-phellandrene	MH	1006	1003	0.3
p-cymene	MH	1027	1025	5.0
cis-sabinene hydrate	MO	1070	1070	0.2
nopinone	MO	1139	1138	0.3
trans-pinocarveol	MO	1141	1140	2.3
pinocarvone	MO	1164	1164	1.5
4-terpineol	MO	1179	1177	0.2
myrtenol	MO	1195	1196	1.3
verbenone	MO	1207	1205	0.7
methyl thymol	MO	1234	1235	0.2
(E)-anethole	PP	1284	1284	0.5
trans-pinocarvylacetate	MO	1298	1297	0.2
theaspirane II	AC	1315	1315	0.1
myrtenyl acetate	MO	1327	1327	0.7
δ-elemene	SH	1340	1338	1.4
α-longipinene	SH	1352	1353	0.5
cyclosativene	SH	1369	1368	0.6
α-ylangene	SH	1372	1372	0.4
α-copaene	SH	1377	1376	1.2
geranyl acetate	MO	1383	1381	0.3
β-elemene	SH	1393	1391	2.6
longifolene	SH	1403	1402	0.1
α-gurjunene	SH	1410	1409	0.3
β-caryophyllene	SH	1419	1419	9.2
β-copaene	SH	1430	1431	0.3
γ-elemene	SH	1434	1433	0.6
trans-α-bergamotene	SH	1437	1435	0.7
epi-β-santalene	SH	1449	1448	0.5
α-humulene	SH	1456	1455	1.0
alloaromadendrene	SH	1462	1460	4.4
β-chamigrene	SH	1476	1475	16.2
germacrene D	SH	1481	1481	4.3
β-selinene	SH	1486	1485	1.4
α-selinene	SH	1495	1494	12.8
α-muurolene	SH	1499	1500	0.1
trans-γ-cadinene	SH	1514	1515	5.2
δ-cadinene	SH	1524	1523	1.5
α-cadinene	SH	1538	1539	0.2
germacrene B	SH	1557	1556	0.3
(E)-nerolidol	SO	1564	1564	0.3
germacrene D-4-ol	SO	1575	1576	1.4
spathulenol	SO	1577	1578	1.1
caryophyllene oxide	SO	1582	1581	0.7
geranyl 2-methylbutyrate	MO	1600	1596	0.4
γ-eudesmol	SO	1632	1632	0.4
τ-cadinol	SO	1641	1640	2.6
Monoterpene hydrocarbons				16.9
Oxygenated monoterpenes				8.3
Sesquiterpene hydrocarbons				65.8
Oxygenated sesquiterpenes				6.5
Phenylpropanoids				0.5
Apocarotenoids				0.1
Total identified				98.1

<sup>a</sup> MH: monoterpene hydrocarbons; MO: oxygenated monoterpenes; SH: sesquiterpene hydrocarbons; SO: oxygenated sesquiterpenes; PP: phenylpropanoids; AC: apocarotenoids. <sup>b</sup> Linear retention index on HP-5MS column, experimentally determined using homologous series of C<sub>8</sub>–C<sub>32</sub> alkanes. <sup>c</sup> Linear retention index taken from the NIST library.

The chemical composition of the volatile compounds extracted from fresh plant material by HS-SPME was different from that reported in a previous study by Al-Qudah et al. [10]. Their chemical investigation demonstrated that the essential oil of flowers

was dominated by a mixture of oxygenated sesquiterpenes (61.12%) and sesquiterpene hydrocarbons (34.76%). Moreover, the main constituents were different:  $\alpha$ -cadinol (16.48%), germacra-1(10),5-diene-3,4-diol (14.45%),  $\gamma$ -cadinene (12.03%), and  $\alpha$ -muurolol (9.89%) were the most abundant ones, while, except for  $\gamma$ -cadinene, they were not detected in our study. Nevertheless, the results from the analysis of HS-SPME fractions and EOs obtained by hydro-distillation are not directly comparable due to the completely different sampling methods. For instance, the low temperature maintained during the HS-SPME sampling avoids the thermal degradation of some constituents, conversely to distillation methods. This aspect can partially justify why many of the major constituents identified by Al-Qudah et al. [10] were not found in our extract or were present in low amounts. For the same reasons, our data differ from older ones published by Senatore & Bruno on *P. spinosa* harvested in Southern Italy. Their results show a predominance of oxygenated sesquiterpenoids in the EO (60.2% of the oil). Among the 38 components that were identified, the most abundant were germacra-1(10),5-dien-3,4-diol (18.4%),  $\alpha$ -cadinol (14.1%), 3-acetoxygermacra-1(10),5-dien-4-ol (13.0%),  $\tau$ -cadinol (8.2%) and  $\delta$ -cadinene (5.8%) [19].

Several volatile plant metabolites and their mixtures are known to exert antimicrobial effects. For instance,  $\beta$ -pinene, one of the main volatile compounds from the aerial parts of *P. spinosa*, is highly toxic to *Candida albicans* and several pathogenic bacteria, such as the methicillin-resistant *Staphylococcus aureus* (MRSA), which causes severe infections in humans [20].  $\beta$ -Caryophyllene, a well-known sesquiterpene, is also toxic to MRSA and exerts anti-inflammatory activity via inhibiting different inflammatory mediators such as interleukin 1 $\beta$ , interleukin-6, and tumor necrosis factor- $\alpha$ . It has also analgesic, myorelaxing, sedative, and antidepressive effects [21]. Other terpenes contributing to the volatile profile of *P. spinosa*, such as p-cymene and  $\beta$ -selinene, are known to exert antioxidant, analgesic, anti-inflammatory, and antimicrobial activities [22,23]. Overall, these data suggest that these compounds may be involved in the therapeutic effects of *P. spinosa*, although further studies are needed to verify this hypothesis.

### 3.2. NMR and HPLC-MS<sup>n</sup> Analysis of Secondary Metabolites

HEX, EA, and DCM solvents allowed the extraction of the most abundant lipophilic constituents of the plant material. The respective spectra, reported in Figures 2–5, support the fact that the most abundant compounds in these fractions are lipids. The assignments of the main signals of these lipids are reported in Table 2.

**Table 2.** Main signals detected in lipophilic fractions attributable to fatty acids.

Signal Assignment	ppm
CH <sub>3</sub> FA: All fatty acids except linolenic acid (-CH <sub>2</sub> -CH <sub>3</sub> )	0.93
CH <sub>2</sub> FA: Acyl chains [-(CH <sub>2</sub> ) <sub>n</sub> -]	1.31–1.35
CH <sub>2</sub> FA1: Acyl chains (-CH <sub>2</sub> -CH <sub>2</sub> -COOH)	1.63
CH <sub>2</sub> FA2: Mono- and polyunsaturated fatty acids (-CH <sub>2</sub> -CH=CH-)	2.10
CH <sub>2</sub> COO: Acyl chains in unsaturated fatty acids (-CH <sub>2</sub> -COOH)	2.35
CH <sub>2</sub> FA CH: (-CH=CH-CH <sub>2</sub> -CH=CH-)	2.80
TGL1: Triacylglycerols (-CH <sub>2</sub> -OCO-)	4.23
TGL2: Triacylglycerols (-CH-OCO-)	5.30
US: unsaturated fatty acids (-CH=CH-)	5.40

In the EA fraction, some minor signals could be observed in the aromatic region and suggest the presence of some phenolic derivatives. In particular, two ortho-coupling doublets at  $\delta$  7.73 and 6.40 indicate the presence of derivatives bearing an ortho-para disubstituted ring, as well as the singlet at  $\delta$  6.50. The meta-coupling doublets at  $\delta$  6.13–6.36 could be ascribed to position 3 and positions 6 and 8 of flavonoid derivatives. Moreover, some minor signals at  $\delta$  7.74, 7.34, and 7.19 indicate the presence of phenolic compounds.

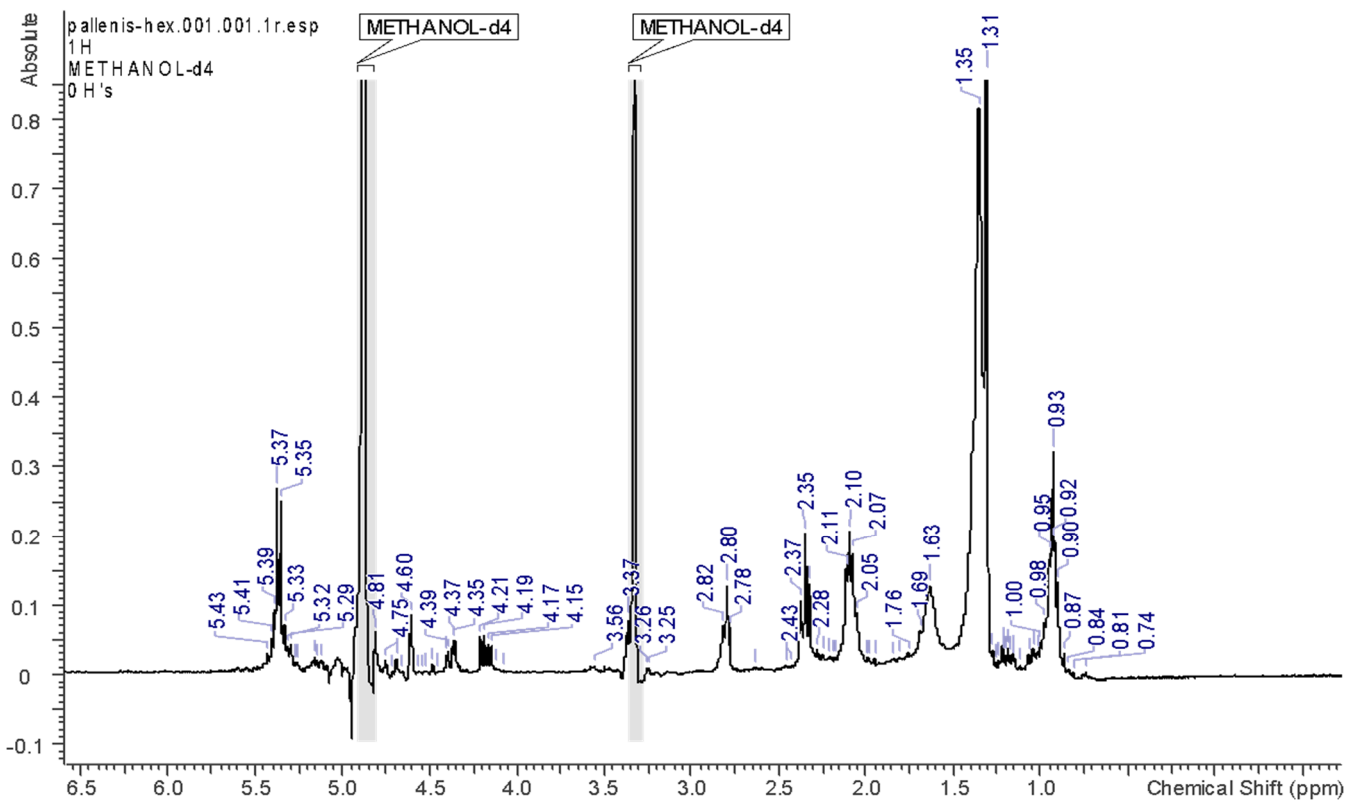


Figure 2. <sup>1</sup>H-NMR spectrum of the HEX fraction.

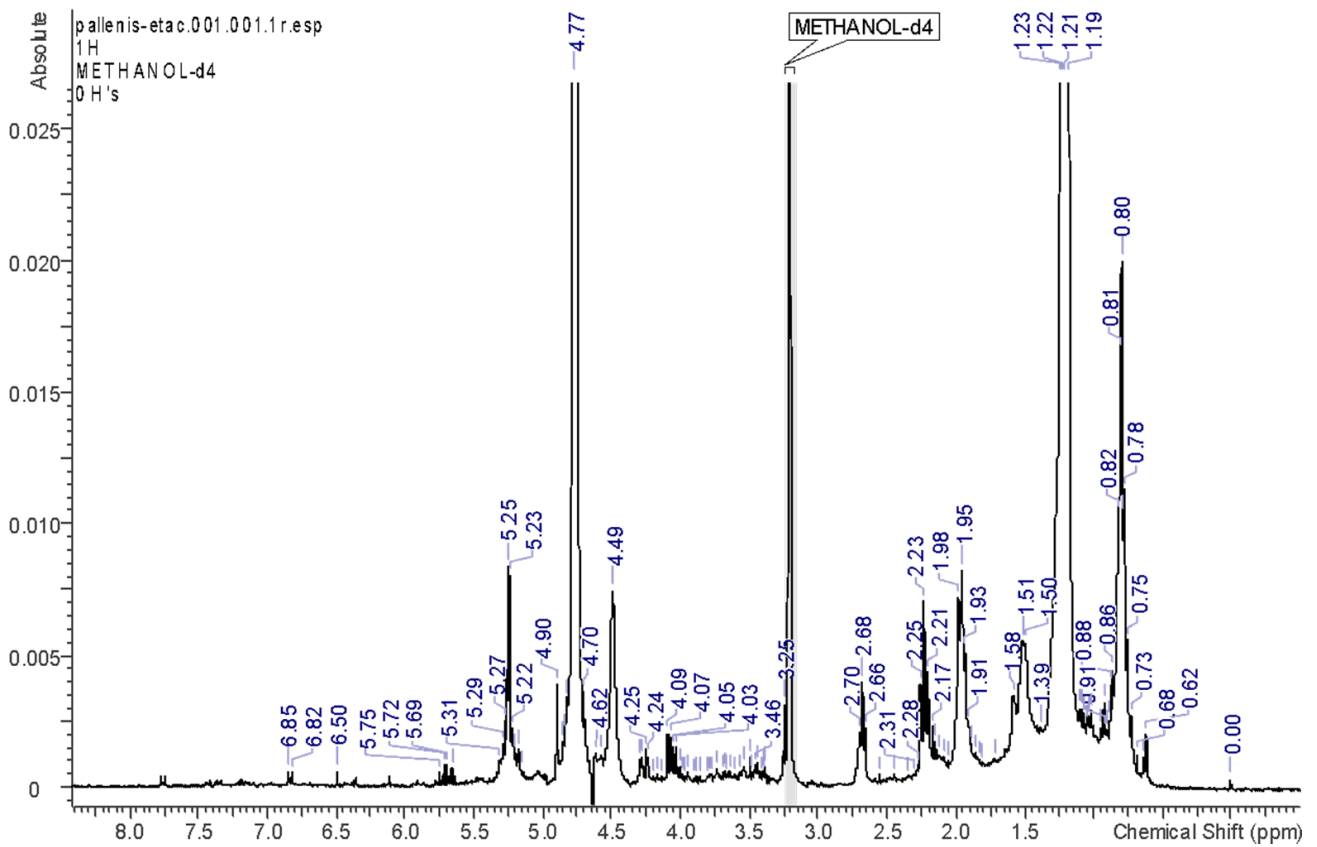


Figure 3. <sup>1</sup>H-NMR spectrum of the EA fraction.

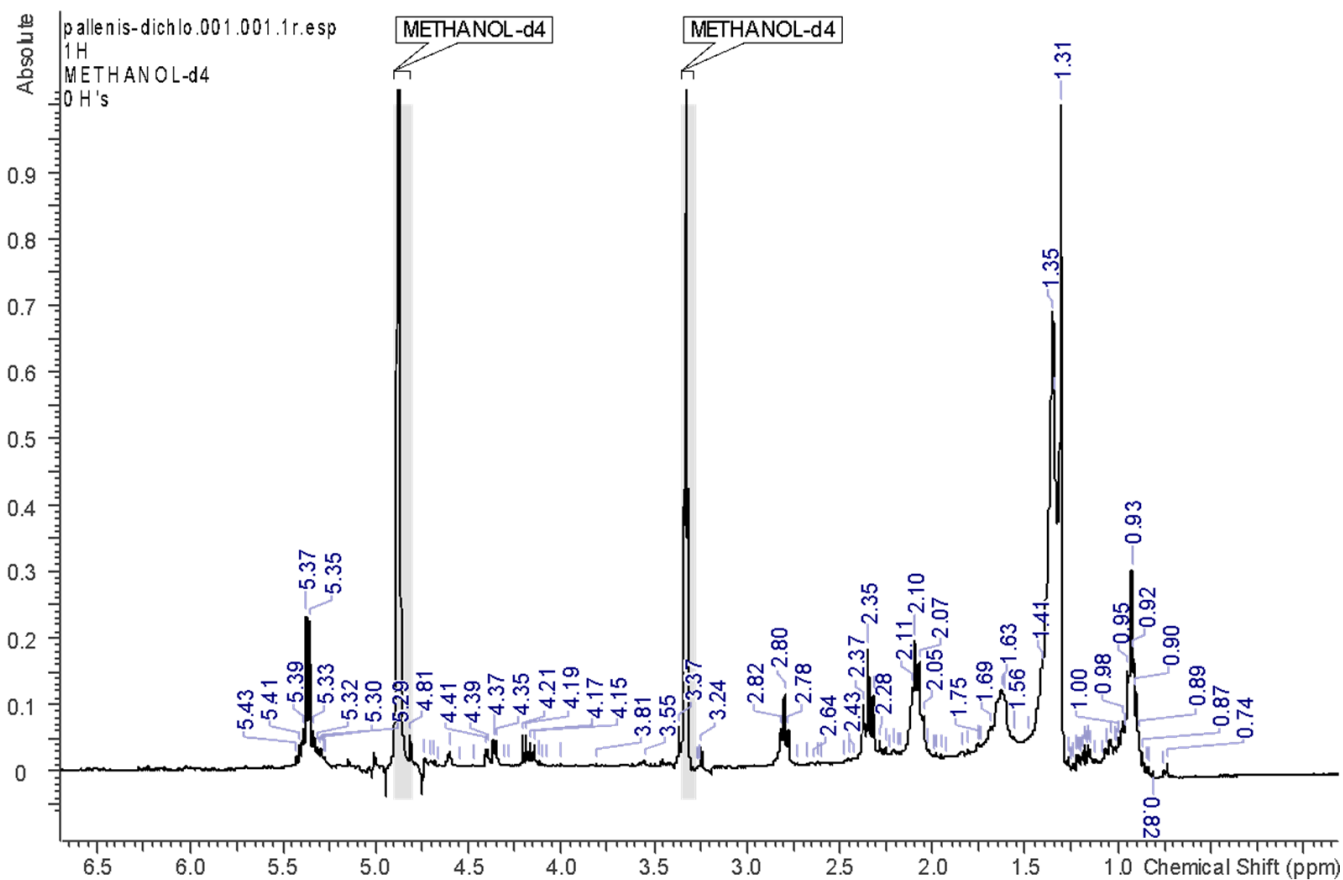


Figure 4. <sup>1</sup>H-NMR spectrum of the DCM fraction.

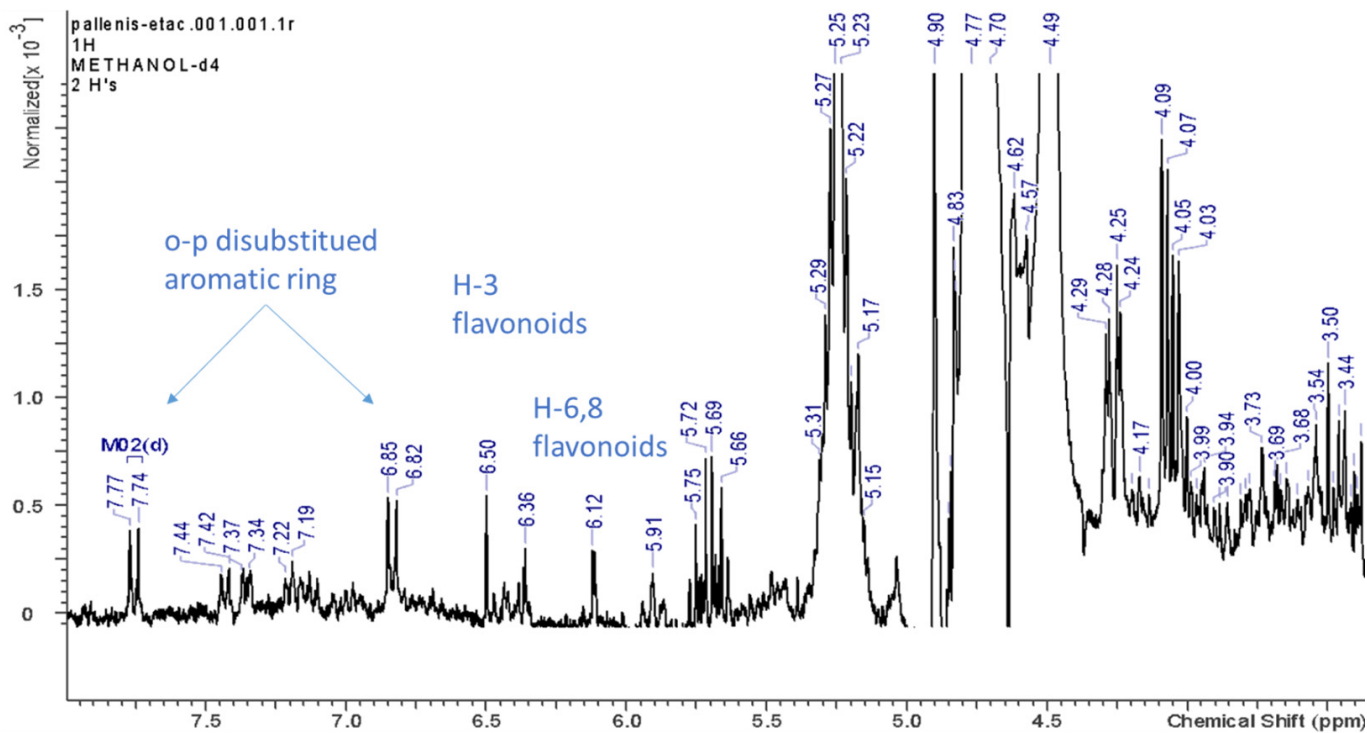
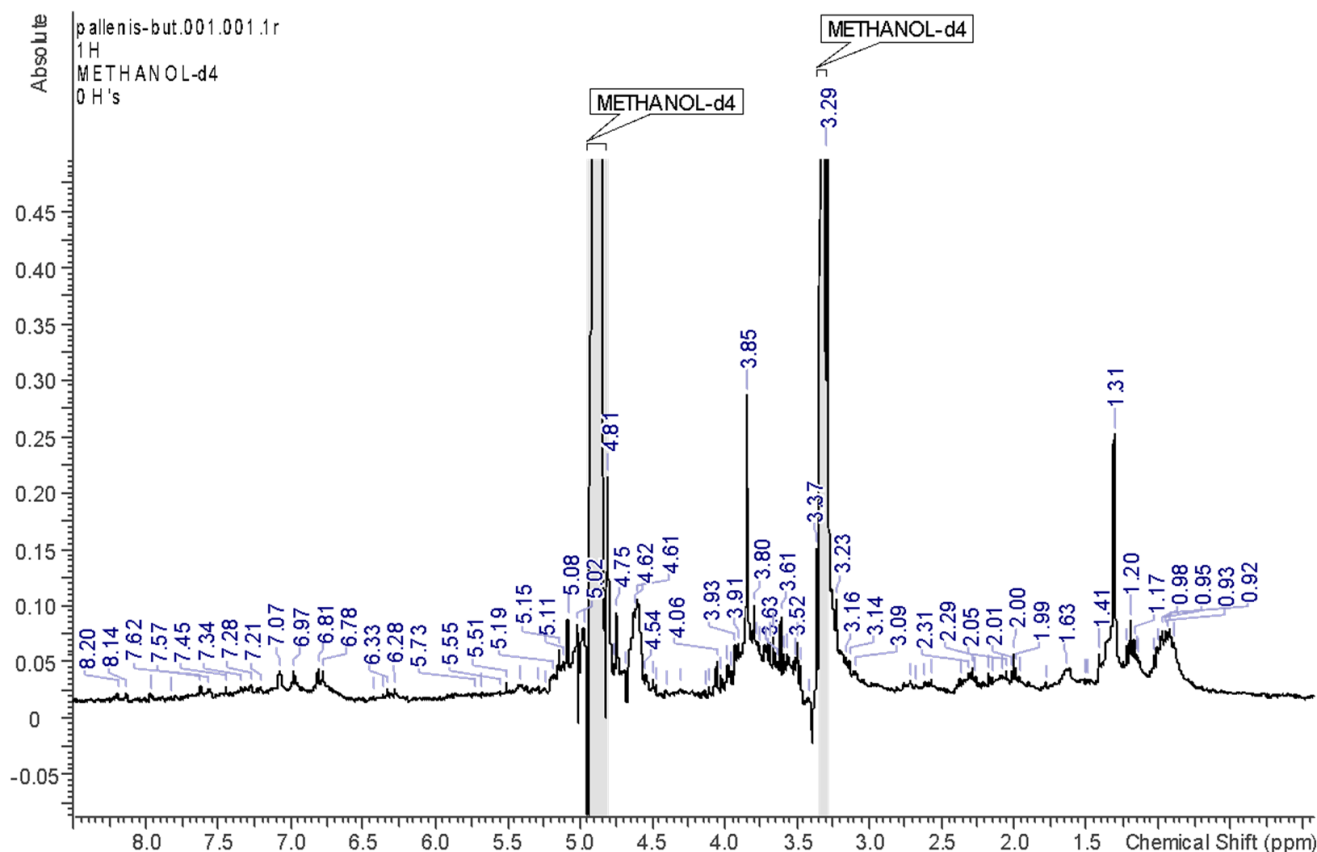


Figure 5. Magnification of the aromatic area in the <sup>1</sup>H-NMR spectrum of the DCM fraction.



On the other hand, the spectrum of the BUT fraction (Figure 6) presents enlarged signals suggesting the presence of several polysaccharides. Nevertheless, some resonances can be useful for a screening of the main compounds. Broad signals in the aliphatic part can be ascribed to the non-oxygenated positions of quinic acid moieties, as well as several aromatic signals supporting the presence of phenols.



**Figure 6.**  $^1\text{H-NMR}$  spectrum of the BUT fraction.

For an exhaustive characterization of phenolic compounds, the chemical composition of the crude methanol extract was explored further by HPLC-MS<sup>n</sup>. As shown in Table 3 and Figure 7, the analysis allowed us to identify 23 compounds as polyphenols. These constituents accounted for 7.26 mg/g of dried plant material, and phenolic acids were the most representative (5.83 mg/g). Tricaffeoylhexaric acid (1.76 mg/g DW), tetracaffeoylhexaric acid (1.41 mg/g DW), 3,5-dicaffeoylquinic acid (1.04 mg/g DW), caffeoyl dihexoside (0.35 mg/g DW), and chlorogenic acid (0.29 mg/g DW) were identified as the most abundant ones. Their MS<sup>n</sup> spectra are shown in Figures S1–S5 of the Supplementary Materials. Among the flavonoids, the most abundant ones were tricetin (0.24 mg/g DW) and its 7-glucoside (0.26 mg/g DW), rutin (0.22 mg/g DW), apigenin (0.18 mg/g DW), and patuletin galactoside (0.15 mg/g DW).

In a paper dated 1992, Ahmed et al. reported the identification of 11 flavonoids in the aerial parts of *P. spinosa* harvested in Egypt. Patuletin 3-O-galactoside and tricetin 7-O-glucoside, also detected in our study, were two of these, together with other tricetin, patuletin, and quercetin glycosides [13]. More recently, Khettaf & Dridi reported the isolation of tricetin 7-O-glucoside from the aerial parts of *P. spinosa* harvested in Algeria, together with tricetin, quercetin, patuletin 7-galactopyranoside, and patuletin-3-O- $\alpha$ -L-rhamnopyranosyl (1–6)- $\beta$ -D-galactopyranoside [24]. A more exhaustive characterization of phenols from the aerial parts of Algerian *P. spinosa* has been presented by Amrani-Allalou et al. In their work, the authors reported 13 compounds that, in accordance with our results, were classified into flavonoids and phenolic acids [25]. Considering the single compounds, several were differ-

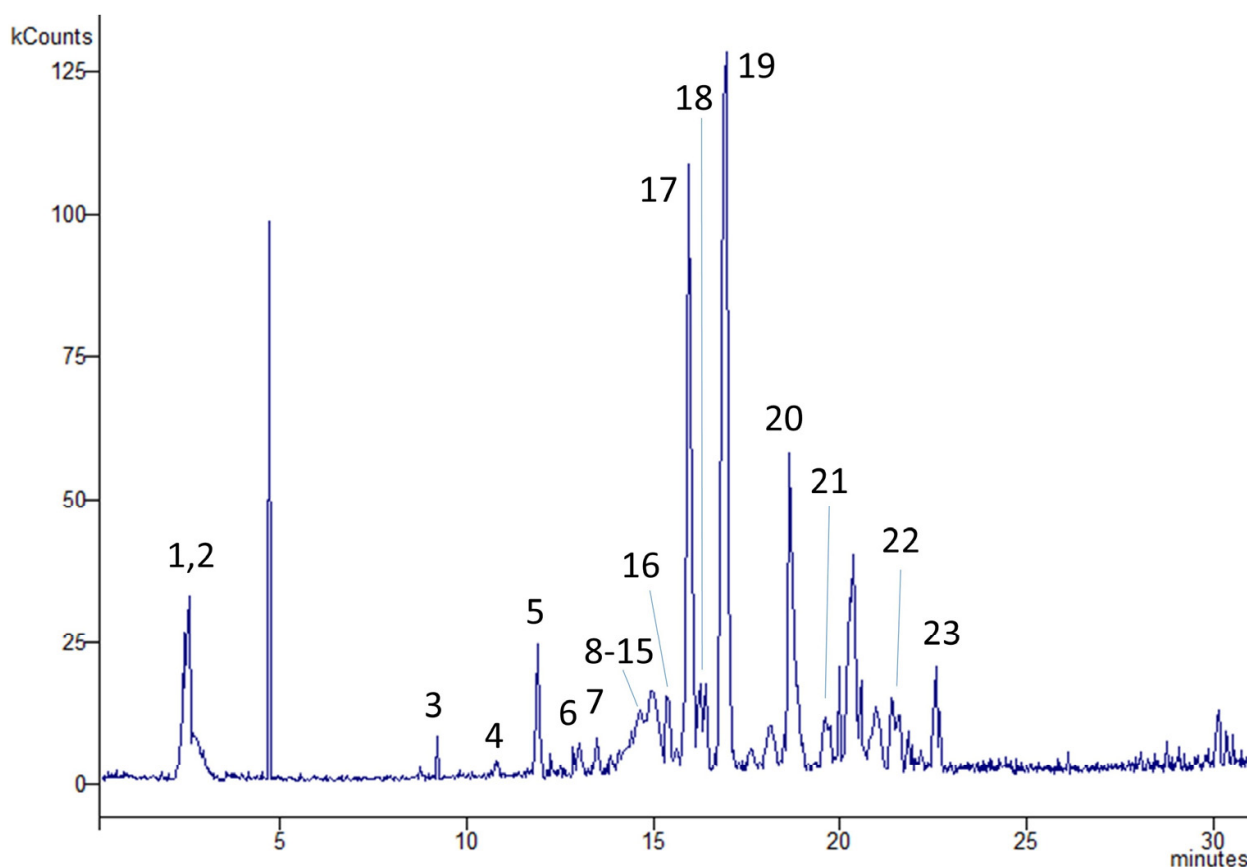
ent from those reported in our work. Among the flavonoids, (epi)catechin, epigallocatechin, phlorizin, naringenin, and apigenin glucoside were not detected in our extract. Conversely, the presence of myricetin hexoside in *P. spinosa* is reported for the first time in our work. Among the phenolic acids, several derivatives of caffeic acid identified in our extract were lacking in the results presented by Amrani-Allalou et al. [25]. These differences can be attributed to several factors. First of all, the different collection sites of the plants generally have a significant influence on the chemical properties of plant specimens. Secondly, the different extraction protocols lead to the enrichment of the extracts in compounds with different physical and chemical properties. Last but not least, the analytical method used for the chemical characterization of extracts.

**Table 3.** Results from the HPLC-MS<sup>n</sup> characterization of secondary metabolites extracted from the aerial parts of *P. spinosa*. Results are reported as means  $\pm$  standard deviation (n = 3). The number associated with each metabolite refers to the corresponding peak in the chromatogram shown in Figure 7.

N.	R.T. (min)	m/z *	MS2 Fragments (m/z)	Tentative Identification	Class **	mg/g DW
1	2.56	539	503 341 281 251 221 179	Caffeoyl dihexoside	PA	0.35 $\pm$ 0.03
2	2.65	193	179 149	Ferulic acid	PA	0.07 $\pm$ 0.00
3	9.33	301	179 151	Quercetin	FV	0.04 $\pm$ 0.00
4	10.85	341	179 161 135	Caffeoyl hexoside	PA	0.05 $\pm$ 0.00
5	11.88	353	191 173 127	Chlorogenic acid	PA	0.29 $\pm$ 0.01
6	13.00	533	371 209 191	Dicaffeoylhexaric acid	PA	0.12 $\pm$ 0.00
7	13.50	533	371 209 191	Dicaffeoylhexaric acid isomer 1	PA	0.11 $\pm$ 0.00
8	13.89	479	317 299 271	Myricetin hexoside	FV	0.09 $\pm$ 0.01
9	13.96	367	191 173	Feruloylquinic acid	PA	0.07 $\pm$ 0.00
10	14.40	533	371 209 191	Dicaffeoylhexaric acid isomer 2	PA	0.16 $\pm$ 0.01
11	14.46	639	331	Patuletin-rhamnopyranosyl-galactopyranoside	FV	0.09 $\pm$ 0.01
12	14.65	533	371 209 191	Dicaffeoylhexaric acid isomer 3	PA	0.08 $\pm$ 0.00
13	14.78	609	301	Rutin	FV	0.22 $\pm$ 0.02
14	15.05	533	371 209 191	Dicaffeoylhexaric acid isomer 4	PA	0.12 $\pm$ 0.00
15	15.18	493	331	Patuletin galactoside	FV	0.15 $\pm$ 0.00
16	15.38	503	341 323 221 179 161	Caffeoyl diglucoside	PA	0.24 $\pm$ 0.02
17	15.95	515	353 191 173	3,5-Dicaffeoylquinic acid	PA	1.04 $\pm$ 0.04
18	16.36	491	329 315 300	Tricin 7-glucoside	FV	0.26 $\pm$ 0.01
19	16.92	695	533 371 209	Tricaffeoylhexaric acid	PA	1.76 $\pm$ 0.10
20	18.66	857	695 533 371 209	Tetracaffeoylhexaric acid	PA	1.41 $\pm$ 0.66
21	19.66	285	270 255	Kaempferol	FV	0.13 $\pm$ 0.00
22	21.56	269	151 117 107	Apigenin	FV	0.18 $\pm$ 0.00
23	22.60	329	314 329 299	Tricin	FV	0.24 $\pm$ 0.01
Total identified						7.26 $\pm$ 0.66
Phenolic acids						5.83 $\pm$ 0.65
Flavonoids						1.42 $\pm$ 0.01

\* m/z values of [M-H]<sup>-</sup> ions are reported, except for metabolite 1, which was identified as [M-H+HCl]<sup>-</sup> adduct. \*\* PA: phenolic acid; FV: flavonoid.

In Algeria, *P. spinosa* is used as a medicinal remedy for several health disorders, especially microbial infections and inflammation. Although a bioactivity screening of extracts was not performed in this study, we could assess the presence of some secondary metabolites whose pharmacological properties have already been studied in vitro and in vivo. Among these are some of the most abundant constituents of the methanol extract of *P. spinosa*, such as chlorogenic acid and 3,5-dicaffeoylquinic acid, whose anti-inflammatory, antioxidant, and antibacterial activities have been reported in vivo. Other bioactive compounds were detected among the flavonoids, such as the well-known apigenin, quercetin, and kaempferol. Moreover, for these, literature data demonstrate their antioxidant, antimicrobial, anti-inflammatory, and antitumor effects in vivo. A summary of these data is reported in Table 4. Overall, these data will be useful as a starting point for further bioactivity assessments on *P. spinosa* from Algeria.



**Figure 7.** Exemplificative chromatogram obtained from the HPLC-MS analysis of the *P. spinosa* methyl extract. Numbers refer to the identified metabolites reported in Table 3. Peaks: (1) caffeoyl dihexoside, (2) ferulic acid, (3) quercetin, (4) caffeoyl hexoside, (5) chlorogenic acid, (6) dicaffeoylhexaric acid, (7) dicaffeoylhexaric acid isomer 1, (8) myricetin hexoside, (9) feruloylquinic acid, (10) dicaffeoylhexaric acid isomer 2, (11) patuletin-rharnnopyranosyl-galactopyranoside, (12) dicaffeoylhexaric acid isomer 3, (13) rutin, (14) dicaffeoylhexaric acid isomer 4, (15) patuletin galactoside, (16) caffeoyl diglucoside, (17) 3,5-dicaffeoylquinic acid, (18) triclin 7-glucoside, (19) tricaffeoylhexaric acid, (20) tetracaffeoylhexaric acid, (21) kaempferol, (22) apigenin, (23) triclin.

**Table 4.** Main bioactivities of polyphenols detected in the methanol crude extract of *P. spinosa*. Information is taken from the literature, and the references for each compound are reported in the Table.

Class	Chemical Constituent	Bioactivities	References
Phenolic acids	Ferulic acid	Anti-hyperlipidemic, antioxidant, and anti-inflammatory effects in vivo	[26]
	Chlorogenic acid	Anti-inflammatory, antioxidant, antibacterial, protection of liver and kidney in vivo	[27]
	3,5-Dicaffeoylquinic acid	Anti-inflammatory and memory enhancer in vivo	[28,29]
Flavonoids	Apigenin	Antitumor, antioxidant, decreases levels of blood glucose, improves cognitive performance in vivo	[30]
	Quercetin	Antioxidant, antimicrobial, anti-inflammatory, antitumor effects in vivo	[31]
	Kaempferol	Antimicrobial and beneficial cardiovascular properties	[32]
	Tricin	Antitumor properties	[33]
	Rutin	Anticancer properties	[34]

#### 4. Conclusions

In this work, we investigated the composition of volatile compounds and non-volatile secondary metabolites of the aerial parts of *P. spinosa* from Algeria. To the best of our knowledge, this study is the first to report an exhaustive chemical characterization of this plant species, considering that, especially for non-volatile metabolites, previous works have described only a few compounds. The results indicate that the volatile profile of Algerian *P. spinosa* is dominated by sesquiterpene hydrocarbons and monoterpene hydrocarbons instead of oxygenated terpenes, conversely to what has been previously reported by other authors. This difference can be related to the different techniques used for extraction and analysis of volatile compounds but also to different times, environmental conditions, and geographical locations of plant harvesting. Regarding non-volatile secondary metabolites, phenolic acids are the most abundant and are represented mainly by caffeic acid derivatives. Flavonoids were also detected, although in low amounts. Nevertheless, some of these are already known to exert beneficial effects in humans; hence, they may be at least co-responsible for the bioactivities associated with *P. spinosa*. In the future, further studies will be required to verify this hypothesis.

Overall, the results of this study represent a contribution to the valorization of understudied plants from Algeria and the preservation of the biodiversity of this region. *P. spinosa*, which is still used as a remedy for several ailments in Algeria, represents a valuable source of phytochemicals such as polyphenols and volatile terpenes. These data may represent a starting point for further rationalization of the medicinal use of this plant species.

**Supplementary Materials:** The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/app131810113/s1>, Figure S1. *m/z* 695: compound identified as tricaffeoylhexaric acid. Figure S2. *m/z* 857: compound identified as tetracaffeoylhexaric acid. Figure S3. *m/z* 515: compound identified as 3,5-dicaffeoylquinic acid. Figure S4. *m/z* 539: compound identified as caffeoyl dihexoside. Figure S5. *m/z* 353: compound identified as chlorogenic acid.

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